

# C-C and C-Heteroatom Bond Forming Reactions on Diiron Bis-Cyclopentadienyl Complexes

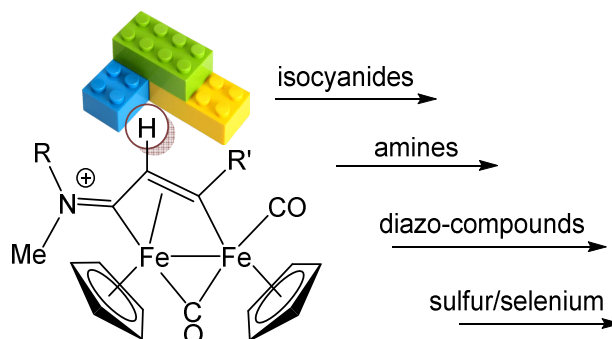
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Dinuclear metal complexes may offer unique reactivity patterns, not attainable with related monometallic species, by leveraging cooperative effects between adjacent metal centers.<sup>1</sup>

Diiron complexes featuring a bridging vinyliminium ligand can be prepared on a multi-gram scale through a broad-scope synthesis, starting from the commercially available and inexpensive  $[\text{Fe}_2\text{Cp}_2(\text{CO})_4]$  ( $\text{Cp} = \eta^5\text{-C}_5\text{H}_5$ ).<sup>2</sup> This class of dinuclear metal complexes is susceptible to a wide variety of functionalization reactions, exploiting the acidity of the vinyliminium moiety and enabling multicomponent assembly strategies under mild conditions (Figure 1). As a result, a diverse array of uncommon yet stable organometallic architectures can be generated.<sup>3</sup> This reactivity will be reviewed, with a focus on the convenient access to a family of multi-substituted selenophenes<sup>4</sup> and the first reproducible synthesis of ferrabenzenes.



**Figure 1.** General structure of diiron vinyliminium complexes, which serve as benchmark compounds for the synthesis of a wide variety of organometallic frameworks, including highly functionalized organic molecules. R = alkyl, aryl; R' = alkyl, aryl, H, carboxylate, SiMe<sub>3</sub>, thiophenyl, pyridyl, ferrocenyl.

## References

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