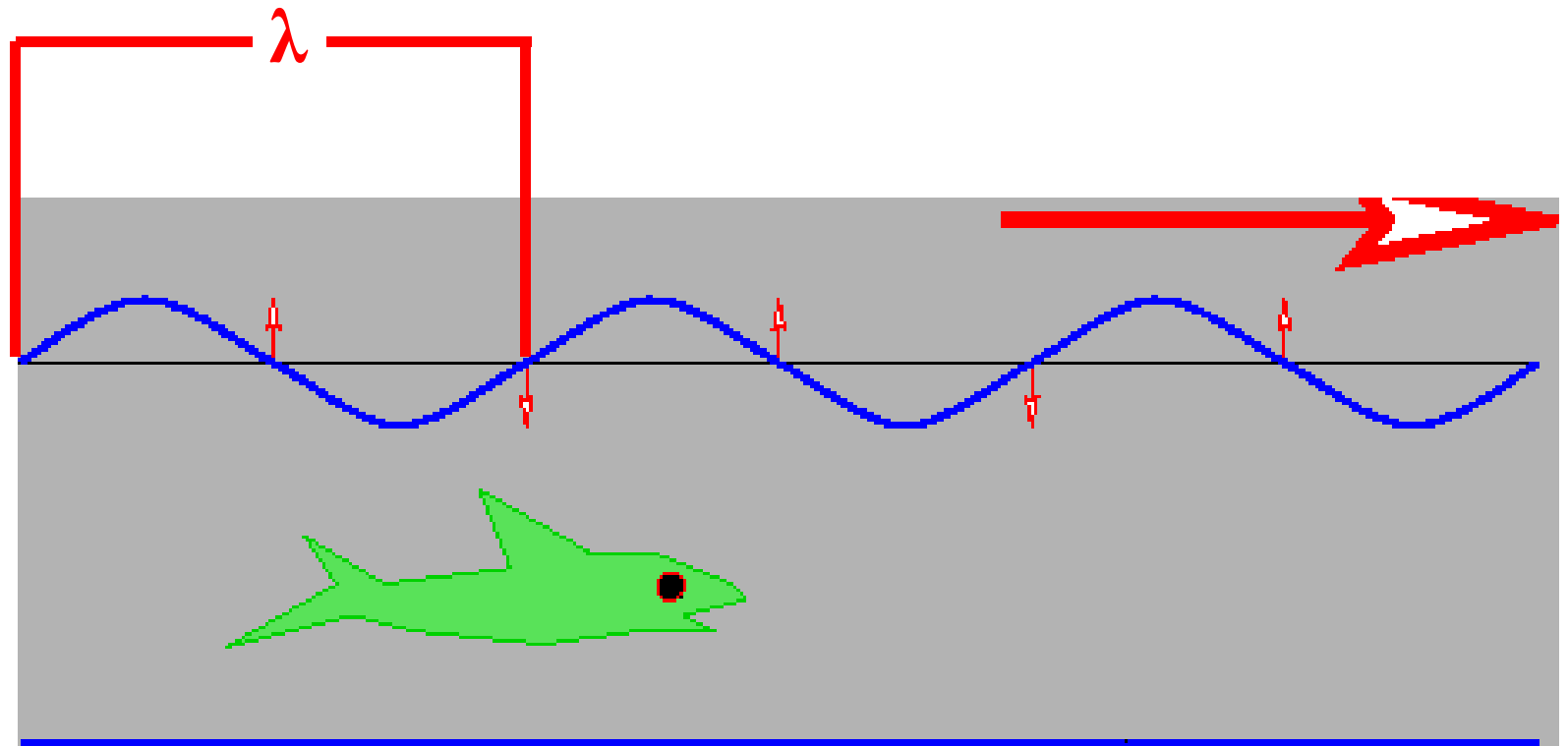
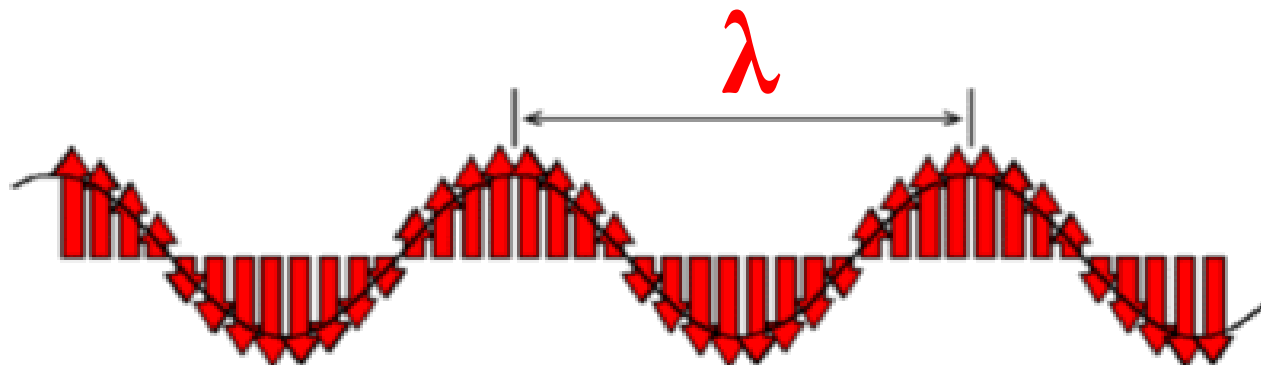


ΑΚΤΙΝΕΣ-Χ

***Ηλεκτρομαγνητική ακτινοβολία
πολύ μικρού μήκους κύματος
και υψηλής ενέργειας***

KYMA

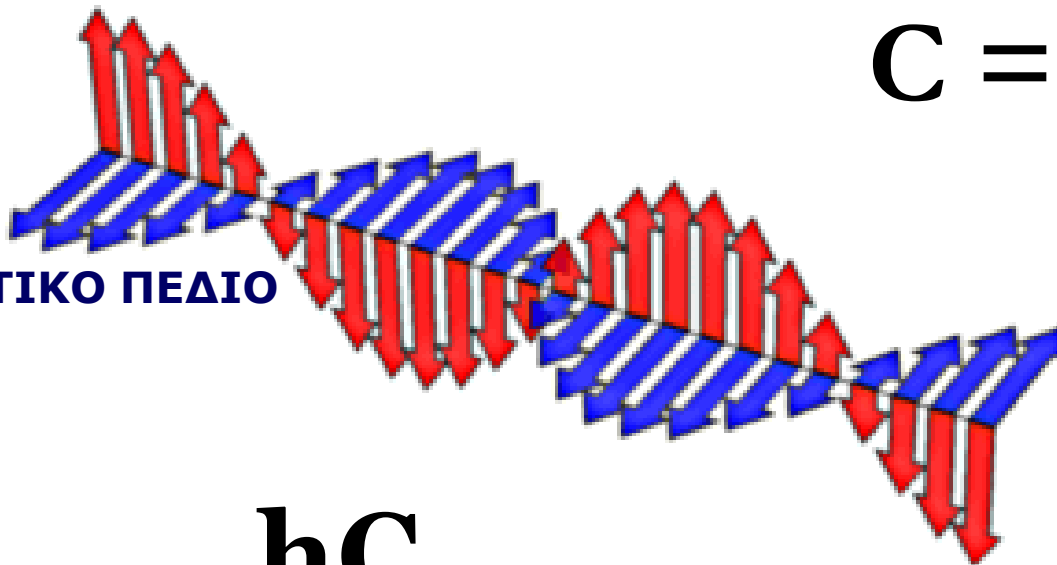




ΗΛΕΚΤΡΙΚΟ ΠΕΔΙΟ

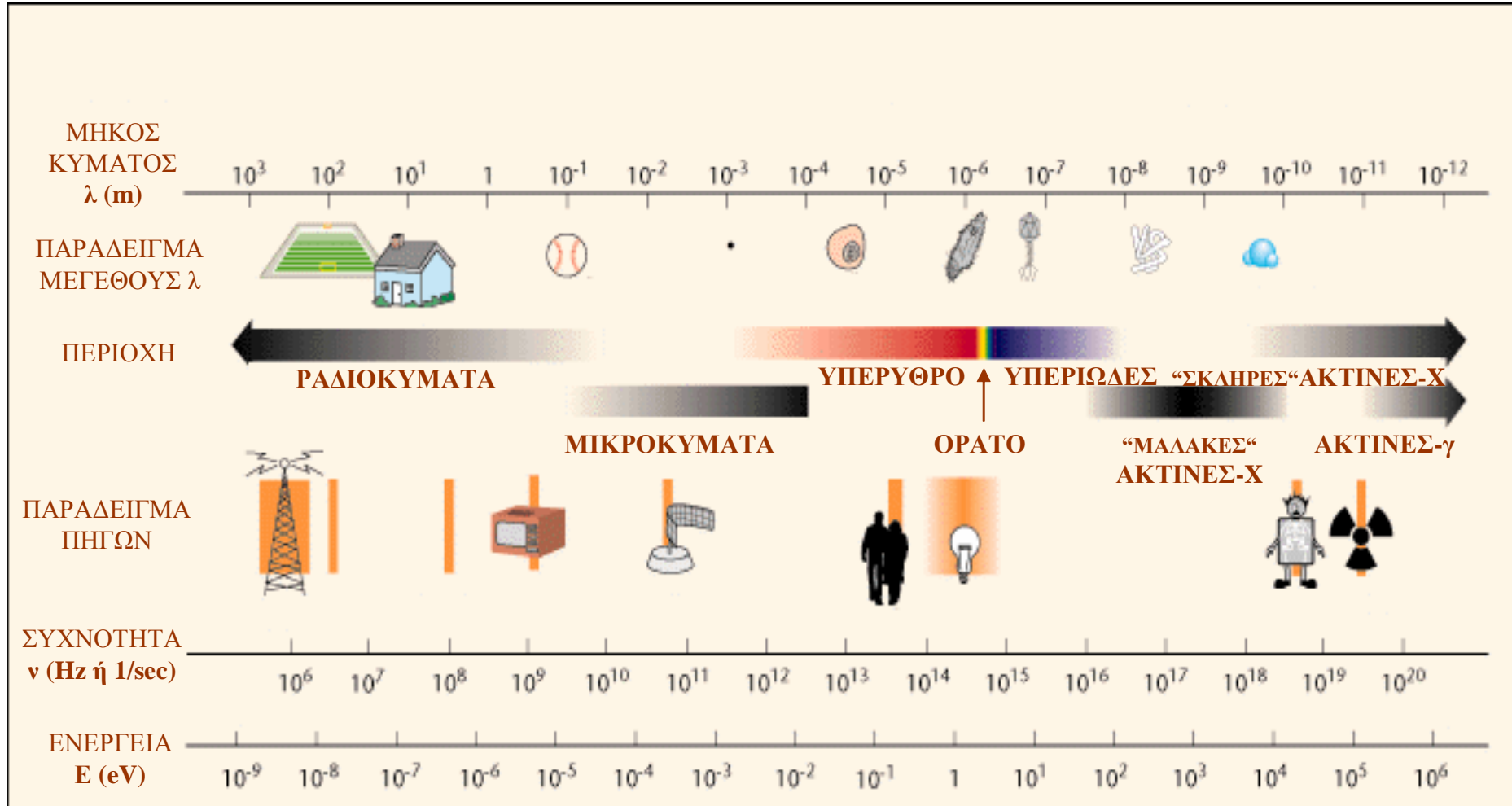
$$C = \nu \lambda$$

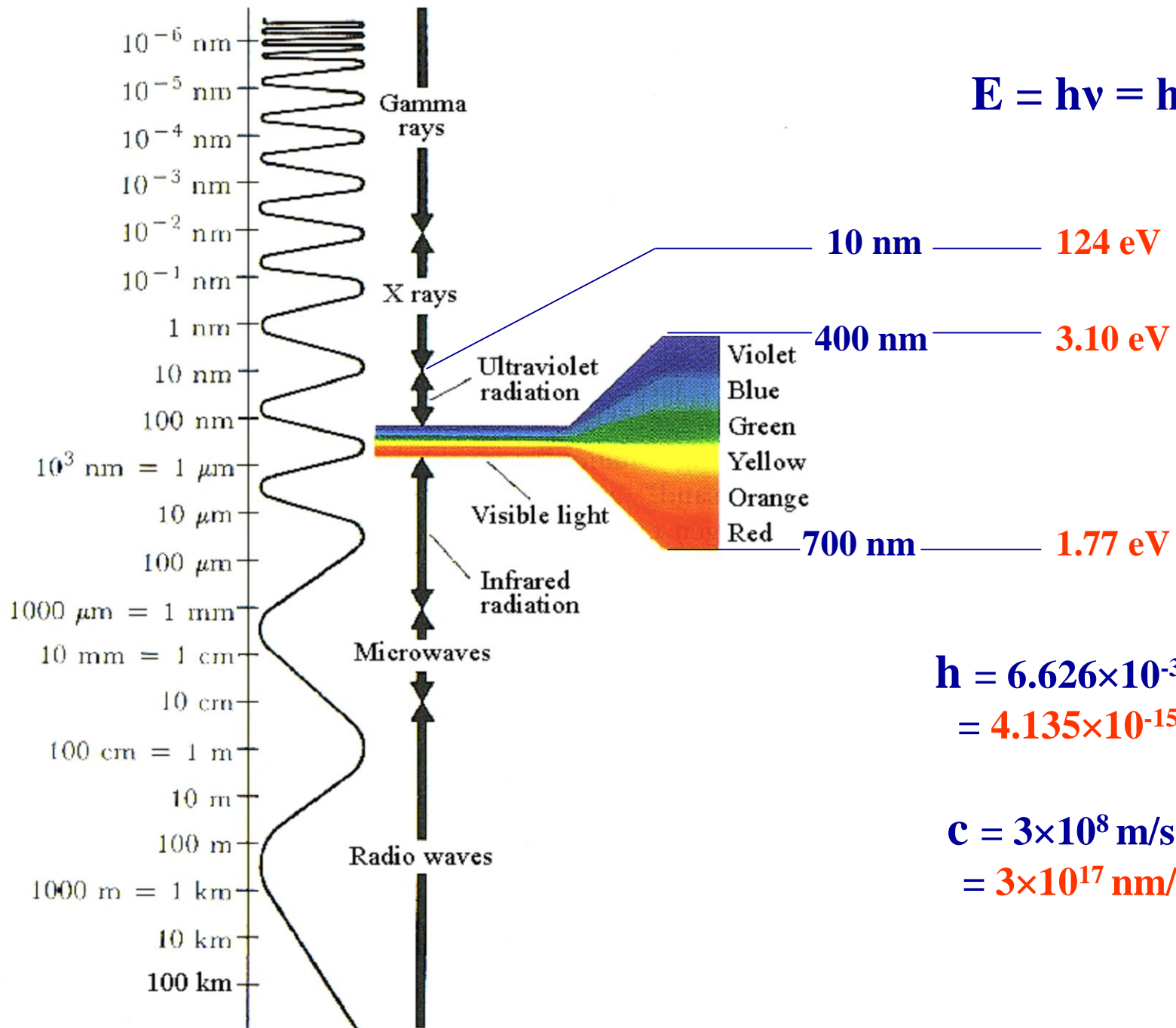
ΜΑΓΝΗΤΙΚΟ ΠΕΔΙΟ



$$E = h\nu = \frac{hc}{\lambda}$$

ΑΚΤΙΝΕΣ-Χ



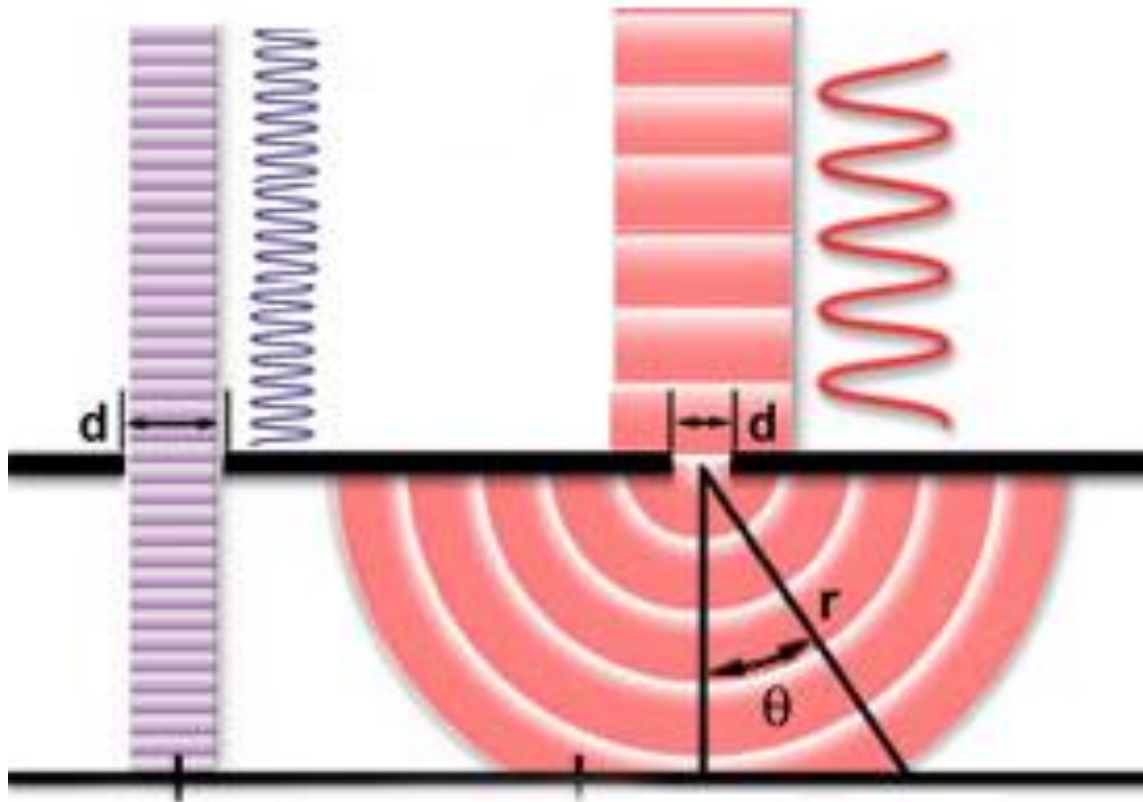


$$E = h\nu = hc/\lambda$$

$$h = 6.626 \times 10^{-34} \text{ Js} = 4.135 \times 10^{-15} \text{ eVs}$$

$$c = 3 \times 10^8 \text{ m/s} = 3 \times 10^{17} \text{ nm/s}$$

ΠΕΡΙΘΛΑΣΗ





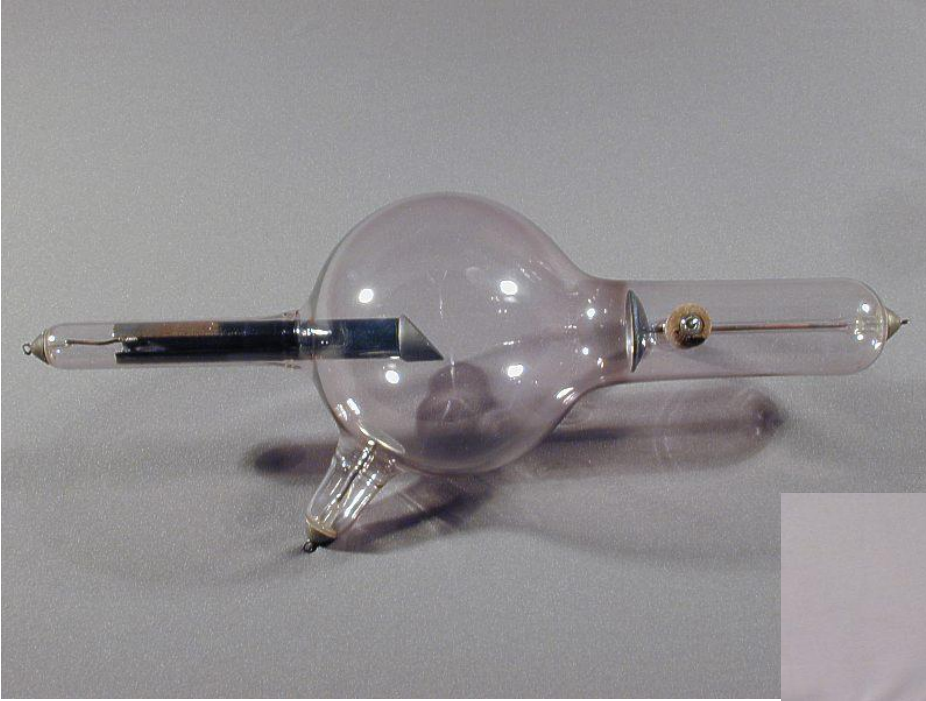
W.C. Röntgen (1845-1923)

ΑΚΤΙΝΕΣ-Χ (RÖNTGEN)

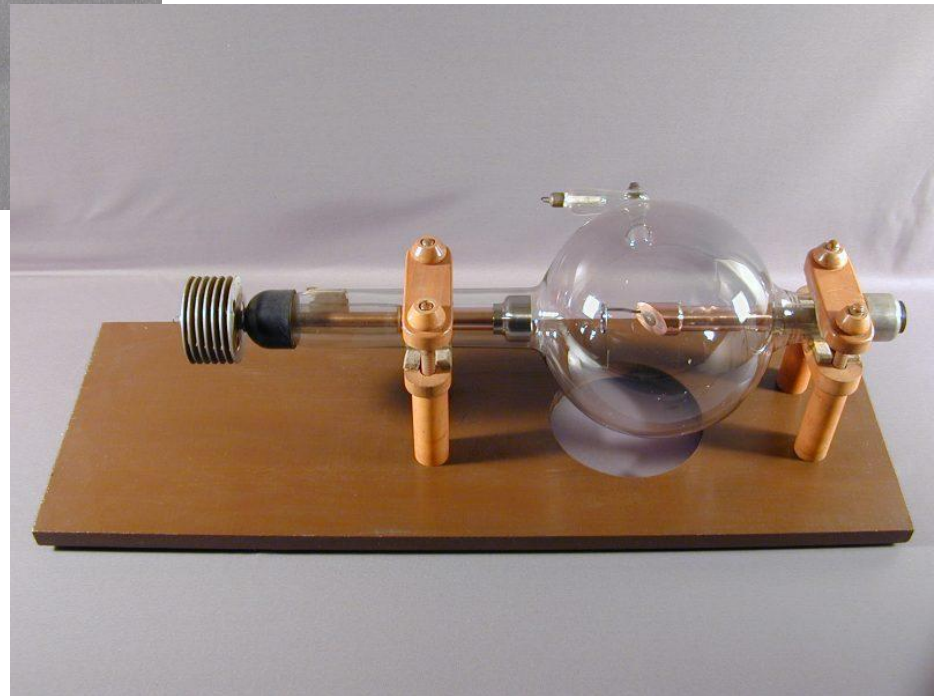
ΒΡΑΒΕΙΟ Nobel ΦΥΣΙΚΗΣ 1901



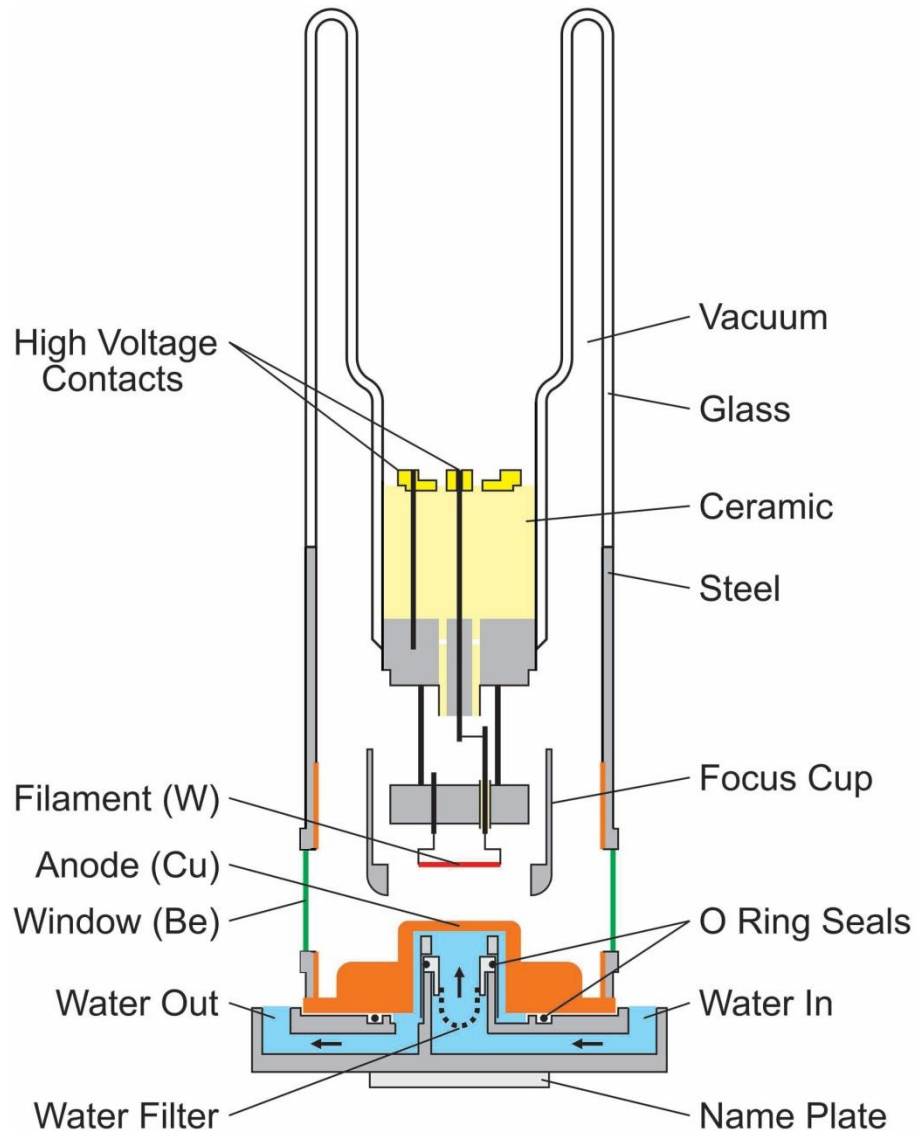
1896



1926



ΚΛΑΣΙΚΟΣ ΓΥΑΛΙΝΟΣ ΣΩΛΗΝΑΣ (ΛΥΧΝΙΑ) ΑΚΤΙΝΩΝ-Χ ΓΙΑ ΠΕΡΙΘΛΑΣΗ (XRD)

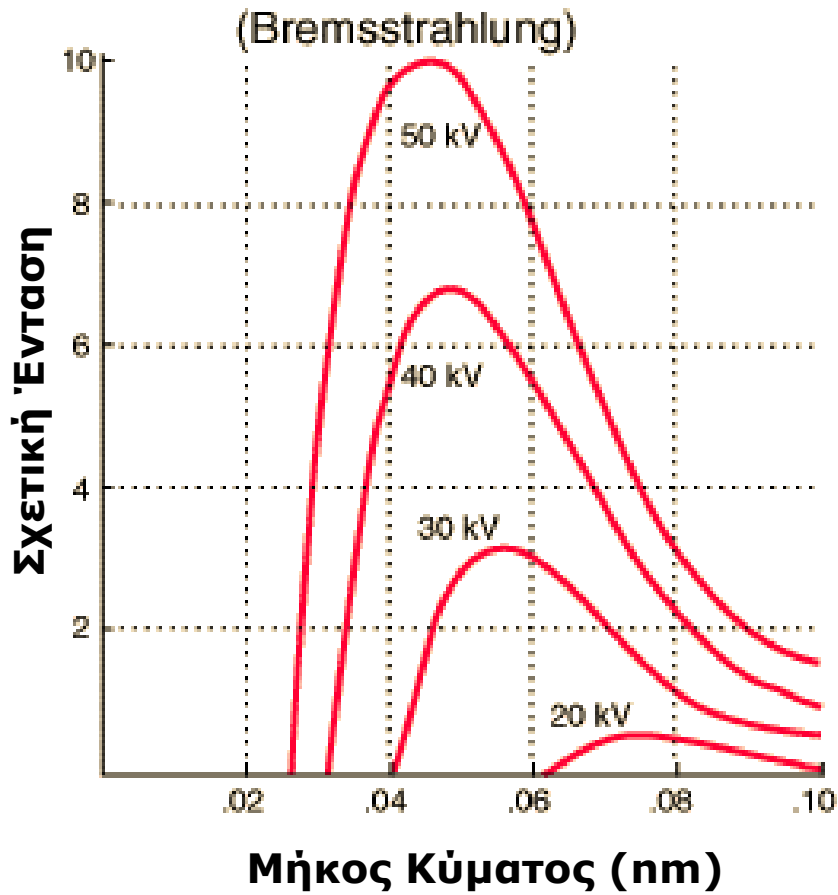


ΣΥΓΧΡΟΝΟΙ ΣΩΛΗΝΕΣ (ΛΥΧΝΙΕΣ) ΑΚΤΙΝΩΝ-Χ ΓΙΑ ΠΕΡΙΘΛΑΣΗ (XRD) ΚΑΙ ΦΘΟΡΙΣΜΟ (XRF)

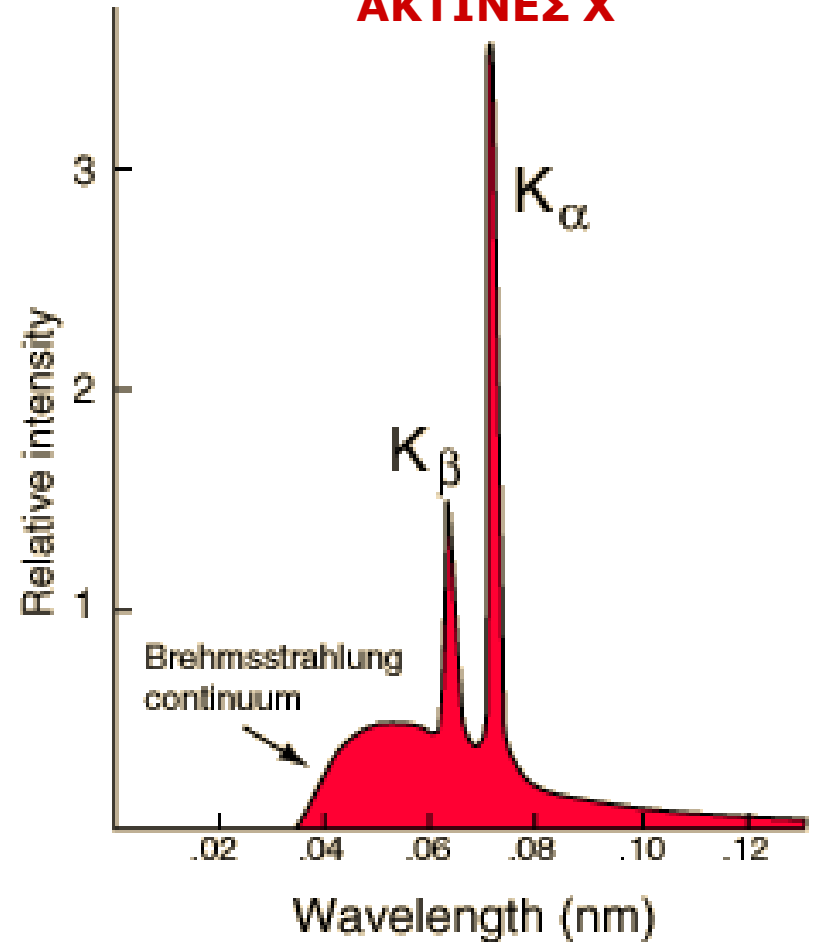


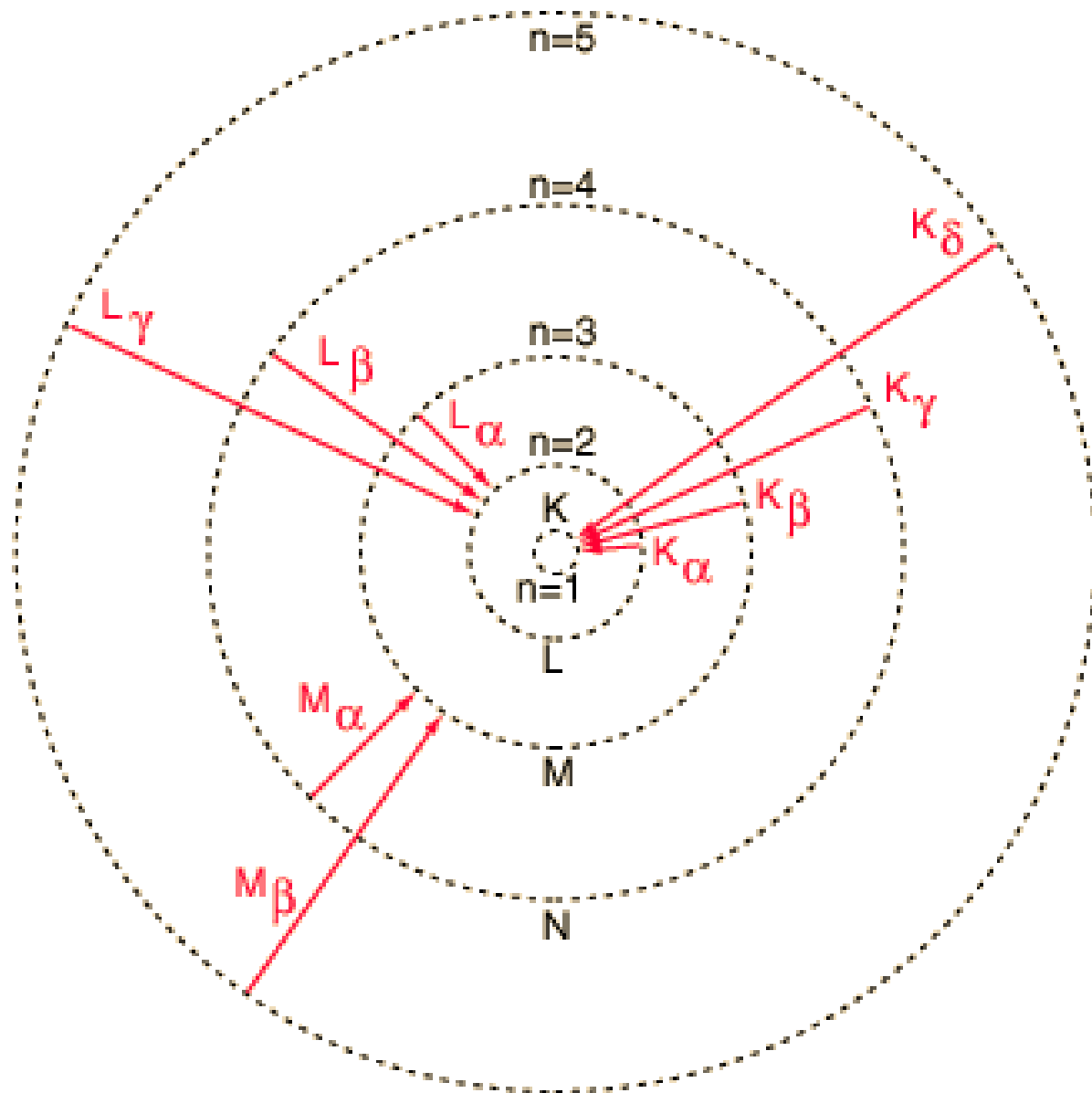
PANalytical

ΣΥΝΕΧΗΣ ΑΚΤΙΝΟΒΟΛΙΑ Χ



ΧΑΡΑΚΤΗΡΙΣΤΙΚΕΣ ΑΚΤΙΝΕΣ Χ







Max von Laue (1879-1959)

ΒΡΑΒΕΙΟ Nobel ΦΥΣΙΚΗΣ 1914

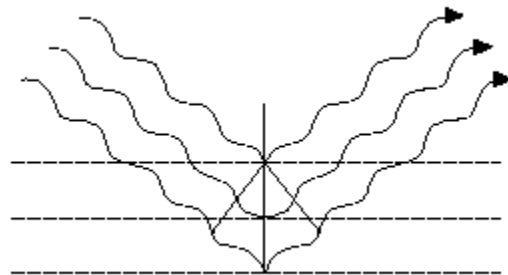


Fig. 1

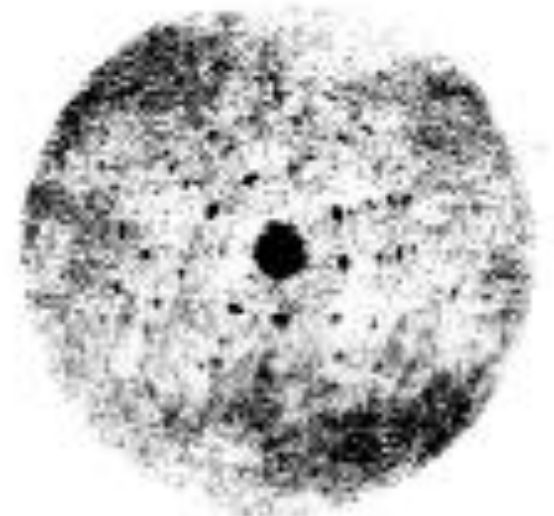


Fig. 2

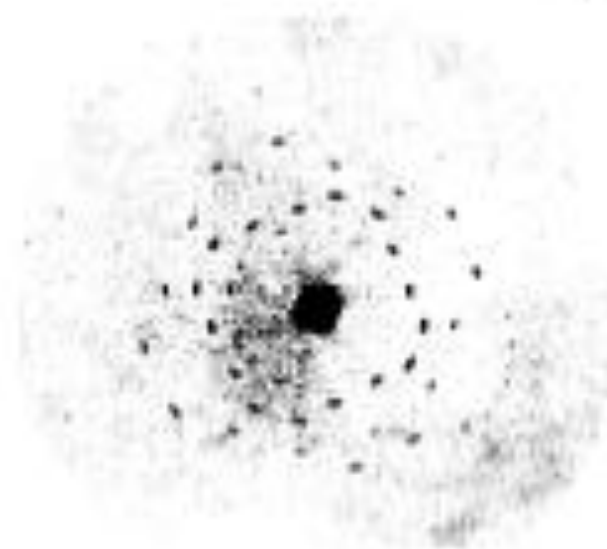
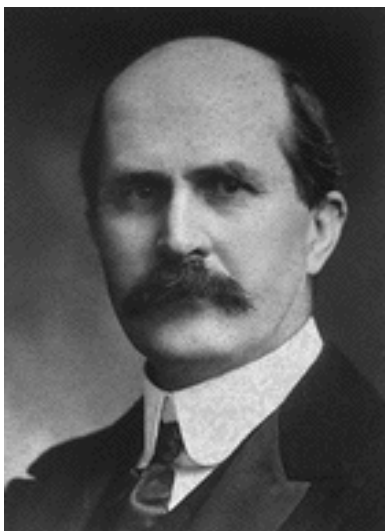


Fig. 3



William Henry Bragg (1862-1942)
William Lawrence Bragg (1890-1971)

ΒΡΑΒΕΙΟ Nobel ΦΥΣΙΚΗΣ 1915

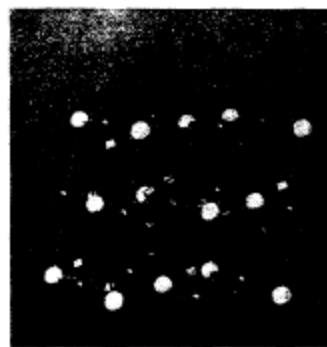


Fig. 6.

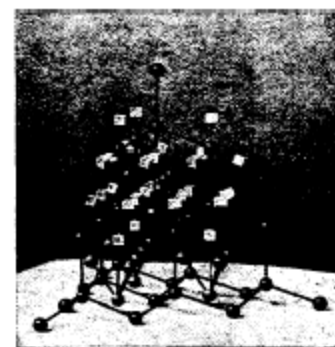
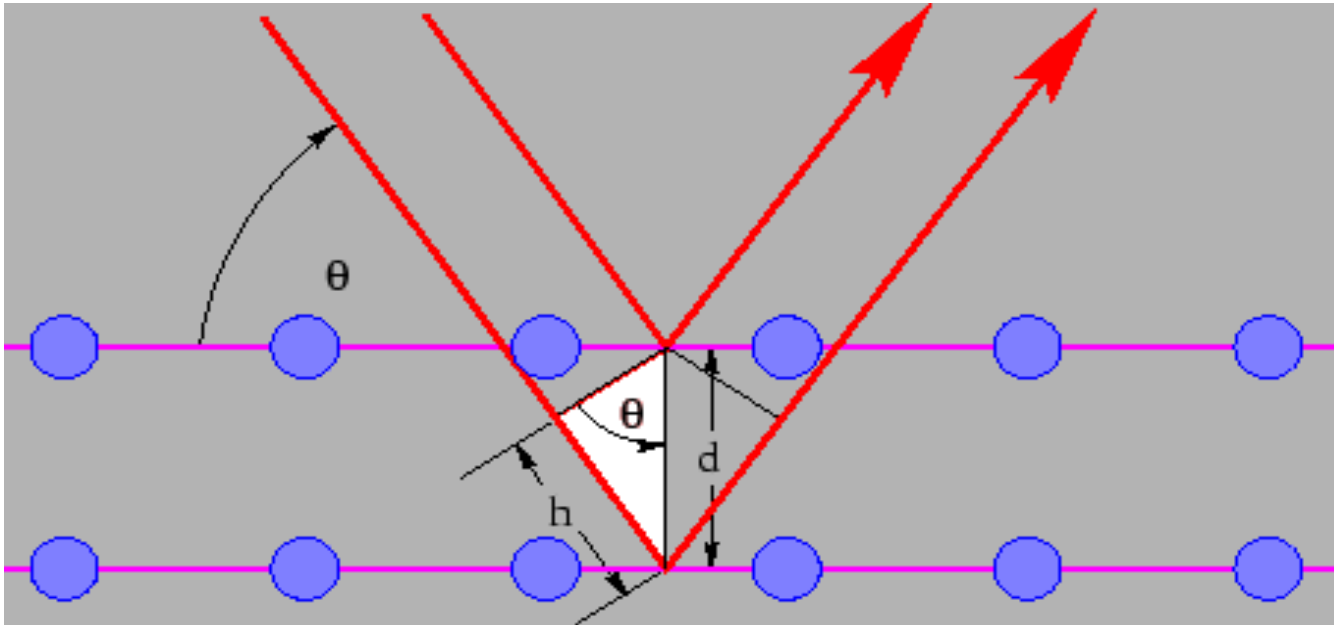


Fig. 7.

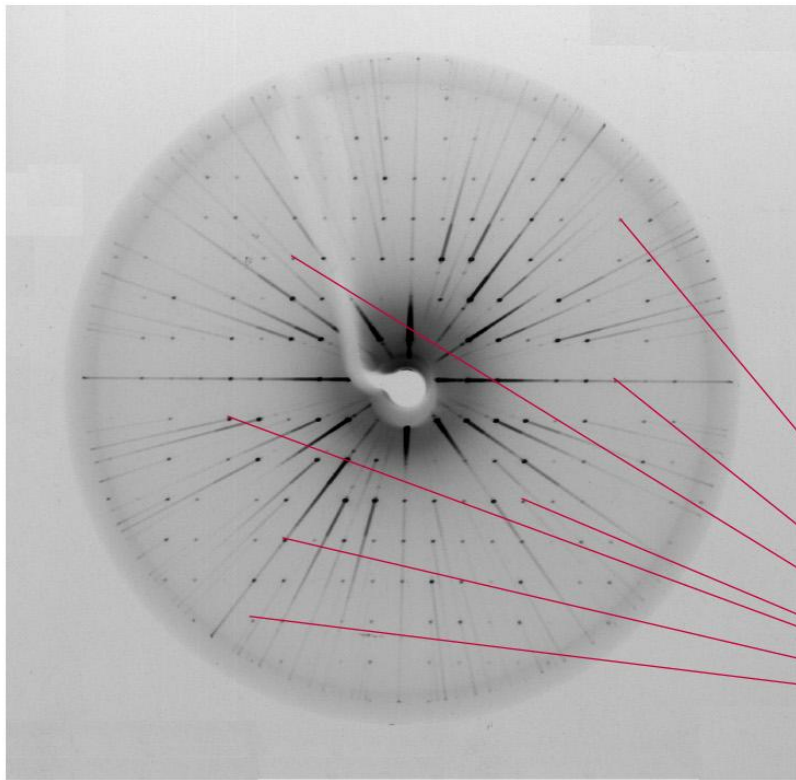
Fig. 6. *Calcite* (CaCO_3). The *white balls* represent calcium; the *black balls*, carbon; and the *cubes*, oxygen.

NΟΜΟΣ Bragg

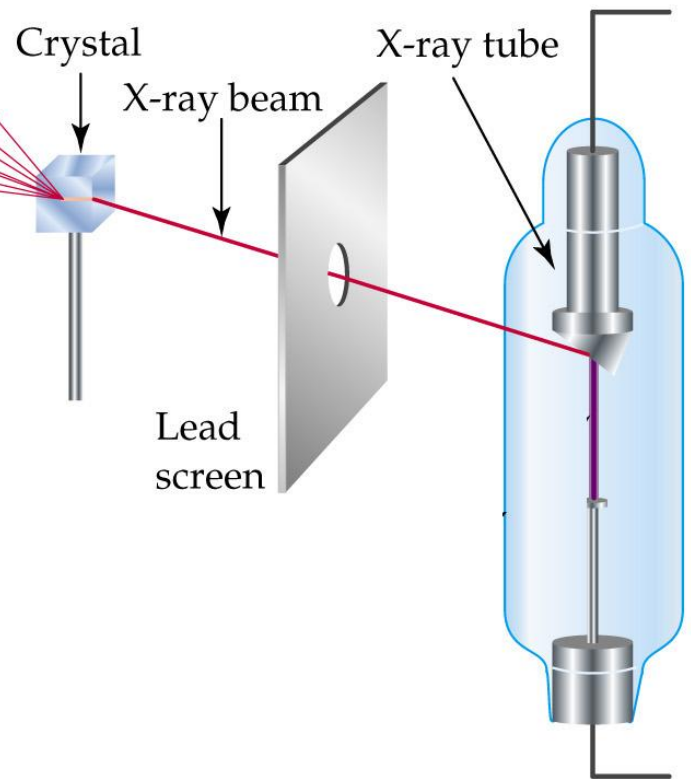
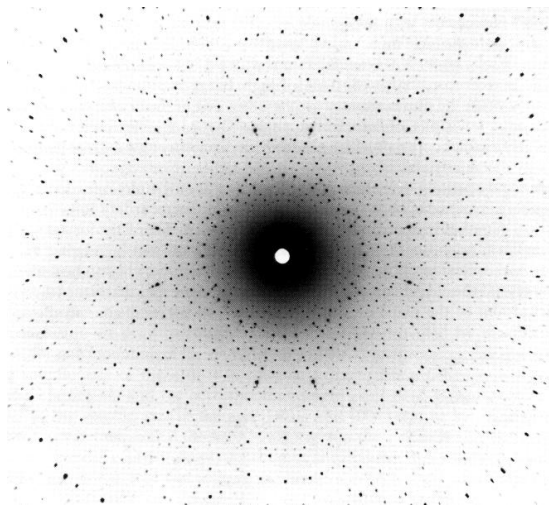


$$n\lambda = 2d\eta\mu\theta$$

ΠΕΡΙΘΛΑΣΗ ΑΚΤΙΝΩΝ-Χ
ΣΕ ΜΟΝΟΚΡΥΣΤΑΛΛΟ
(single-crystal XRD)



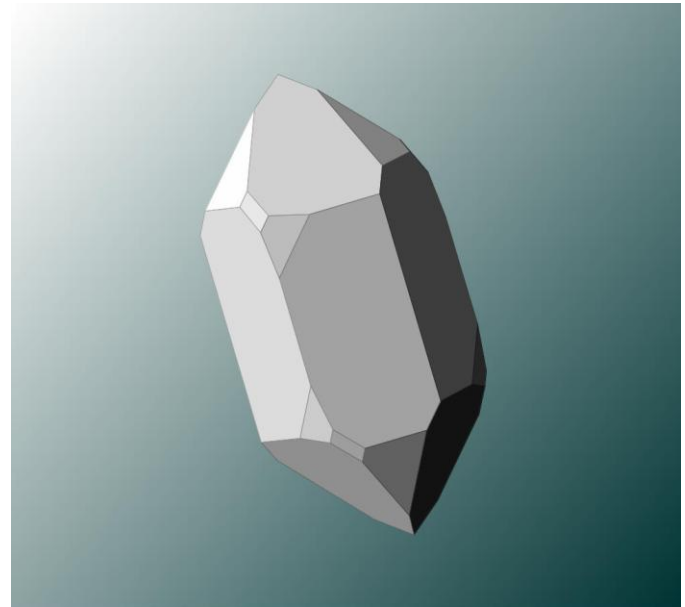
Photographic film



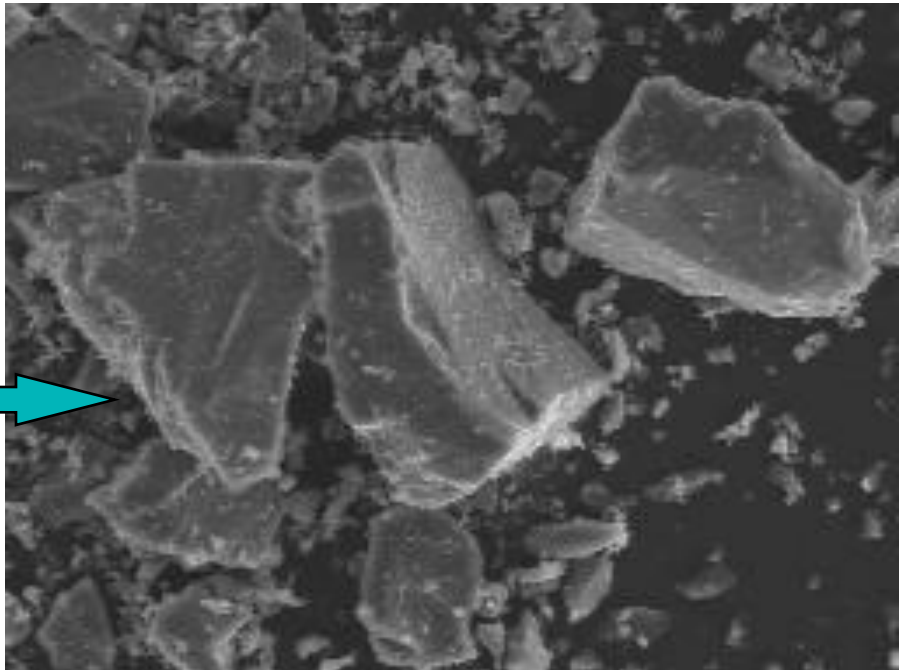
ΠΕΡΙΘΛΑΣΗ ΑΚΤΙΝΩΝ-Χ

ΣΕ ΣΚΟΝΗ

(powder XRD / PXRD)



ΧΑΛΑΖΙΑΣ (SiO₂)





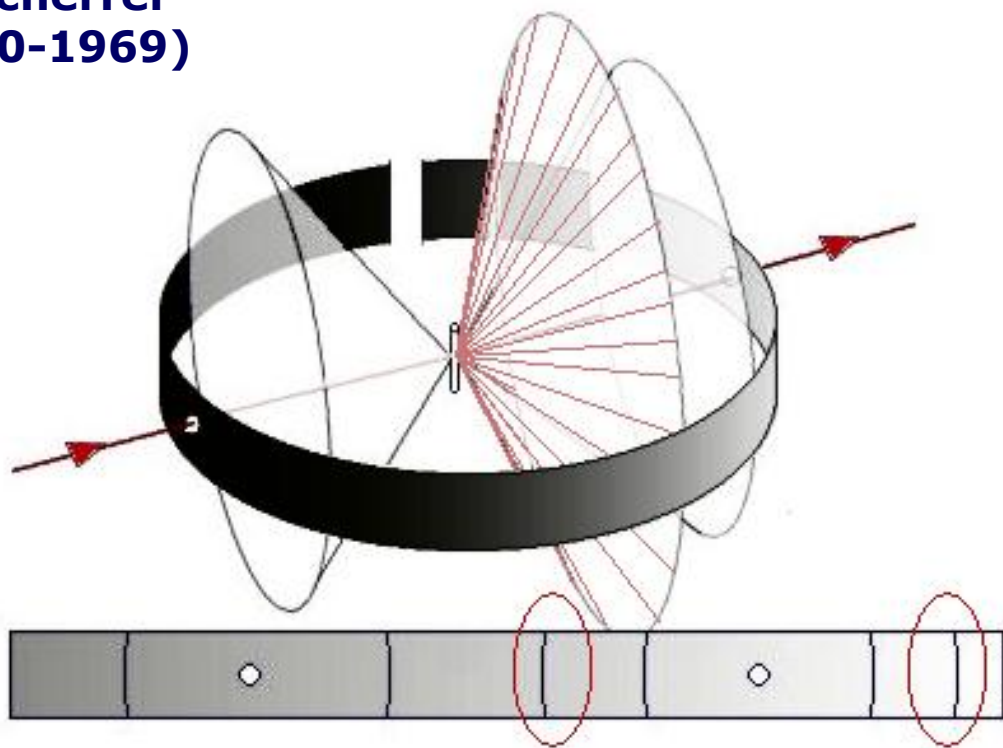
**P.W. Debye
(1884-1966)**

**ΒΡΑΒΕΙΟ Nobel
ΧΗΜΕΙΑΣ 1936**



**P. Scherrer
(1890-1969)**

Μέθοδος Debye- Scherrer





NaCl



Na₂SO₄



Περιθλασίμετρο Ακτίνων-Χ Σκόνης

ΑΝΙΧΝΕΥΤΗΣ ΑΚΤΙΝΩΝ-Χ

X-Rays

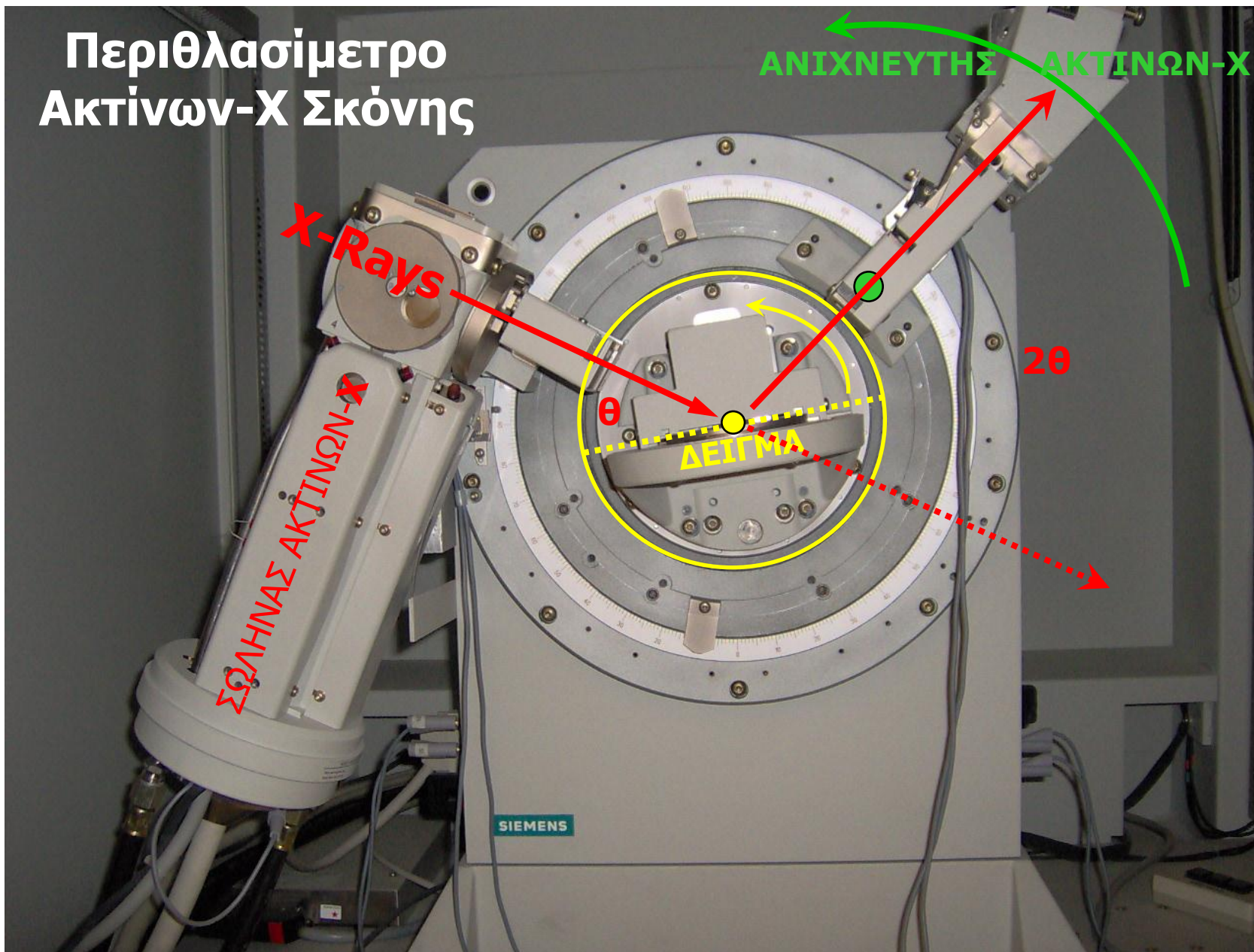
ΣΩΛΗΝΑΣ ΑΚΤΙΝΩΝ-Χ

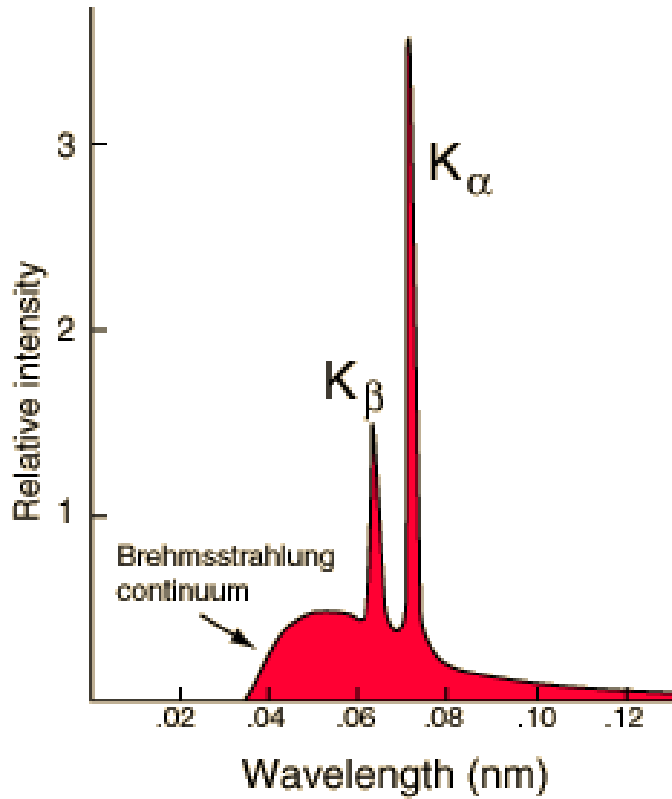
θ

ΔΕΙΓΜΑ

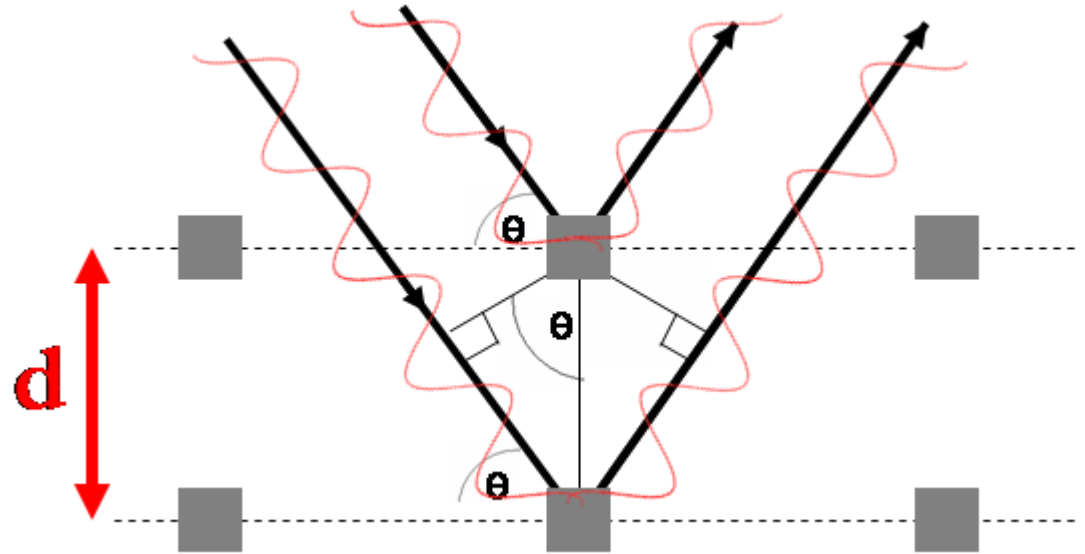
2θ

SIEMENS





Mo K_{α} :
 $\lambda \approx 0.71 \text{ \AA}$

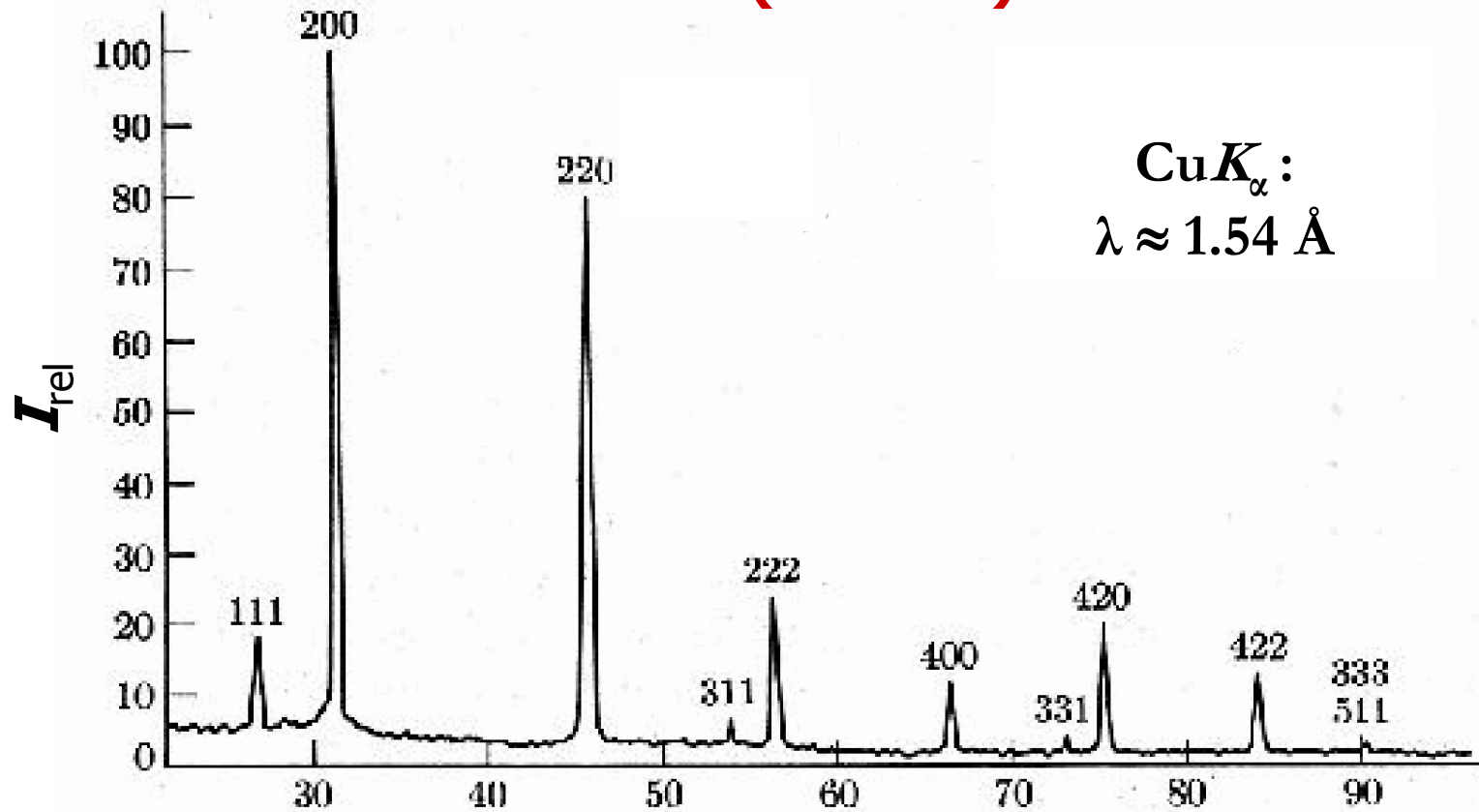


$$n\lambda = 2d \sin \theta$$

ΔΙΑΓΡΑΜΜΑ ΠΕΡΙΘΛΑΣΗΣ ΑΚΤΙΝΩΝ-Χ ΣΚΟΝΗΣ

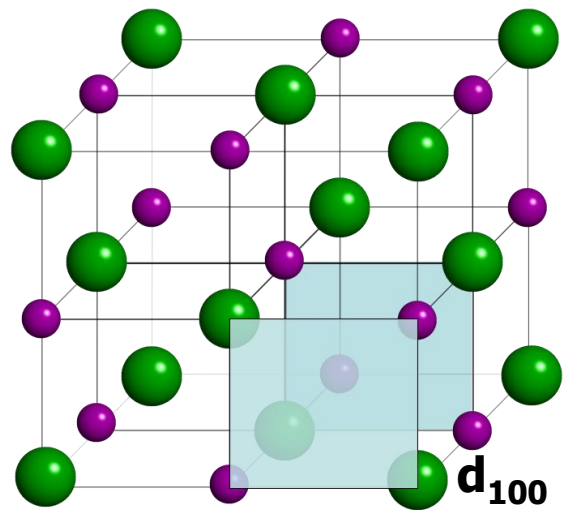
NaCl (ΑΛΙΤΗΣ)

CuK_α :
 $\lambda \approx 1.54 \text{ \AA}$

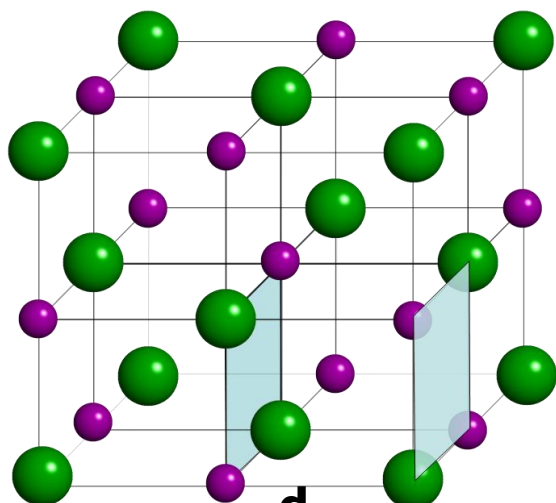


$$I(\text{INTENSITY})_{rel} = \frac{\text{Counts} \times 100}{\text{Counts}_{max}}$$

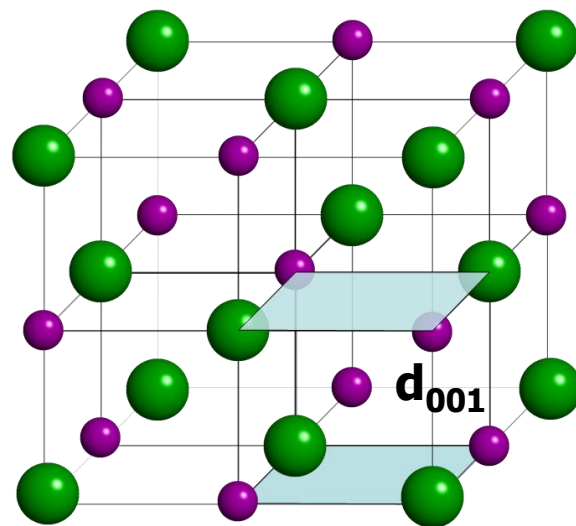
$2\theta(\circ)$



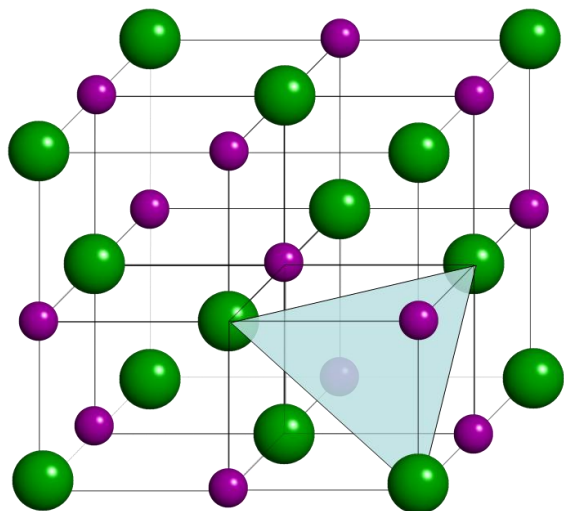
(100)



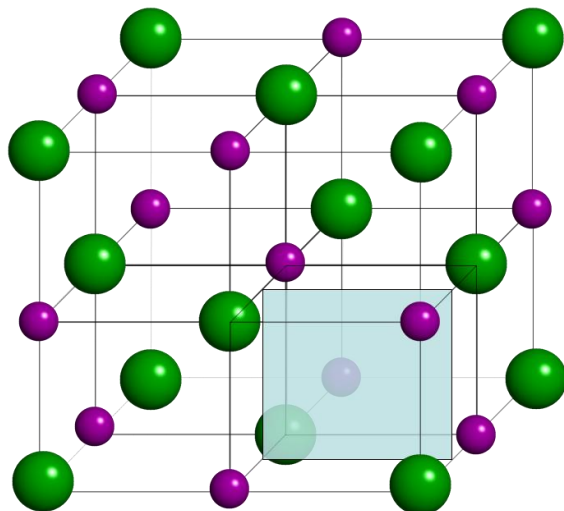
(010)



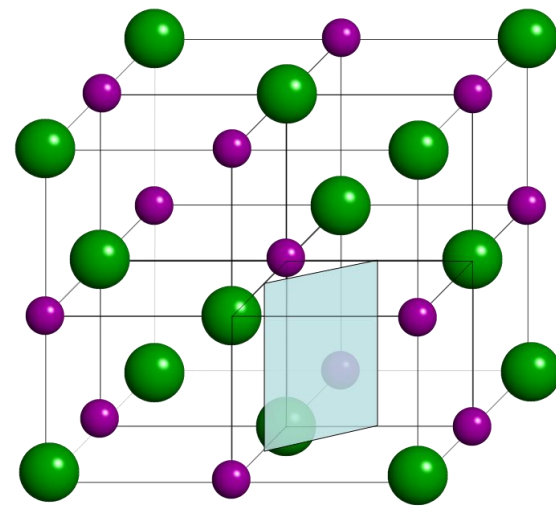
(001)



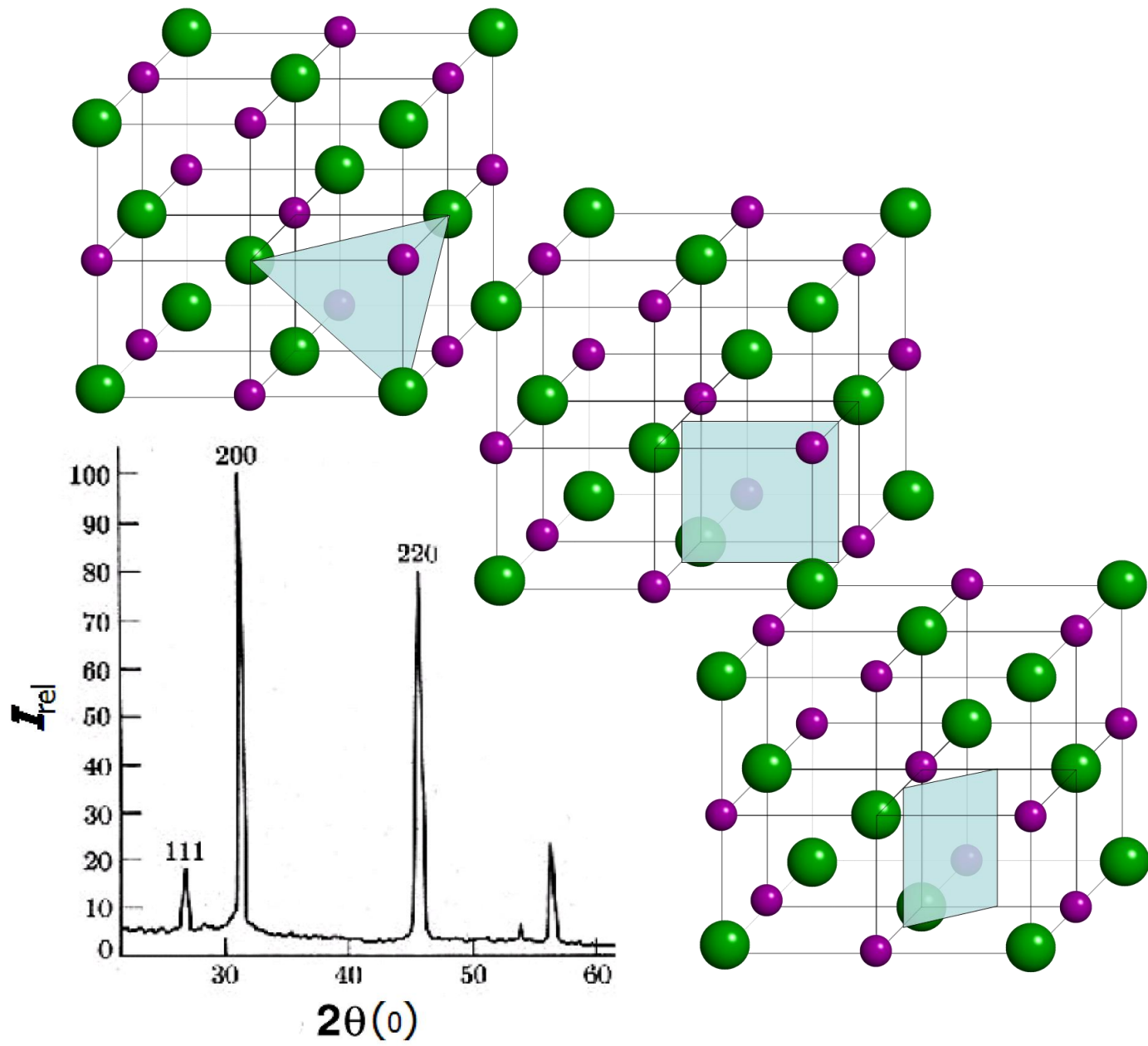
(111)



(200)



(220)



KYBIKO: $a=b=c=a_0$ και $d_{hkl} = a_0/(h^2+k^2+l^2)^{1/2}$

$$\lambda = 2d_{hkl}\sin\theta_{hkl} \Rightarrow \lambda^2 = 4d_{hkl}^2\sin^2\theta_{hkl} \Rightarrow$$

$$\lambda^2 = (4a_0^2/(h^2+k^2+l^2))\sin^2\theta_{hkl} \Rightarrow$$

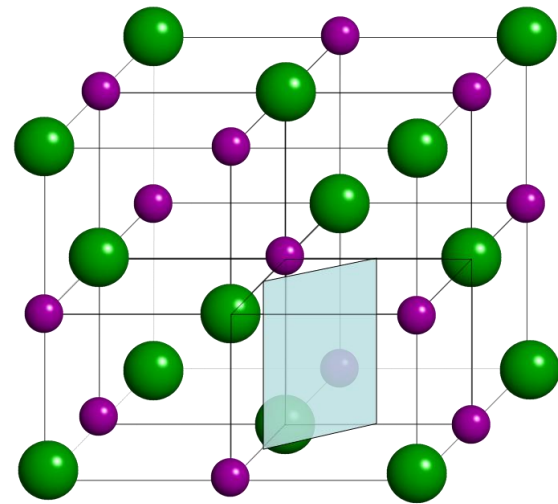
$$\lambda^2/4a_0^2 = \sin^2\theta_{hkl}/(h^2+k^2+l^2) = \sin^2\theta_{hkl}/s$$

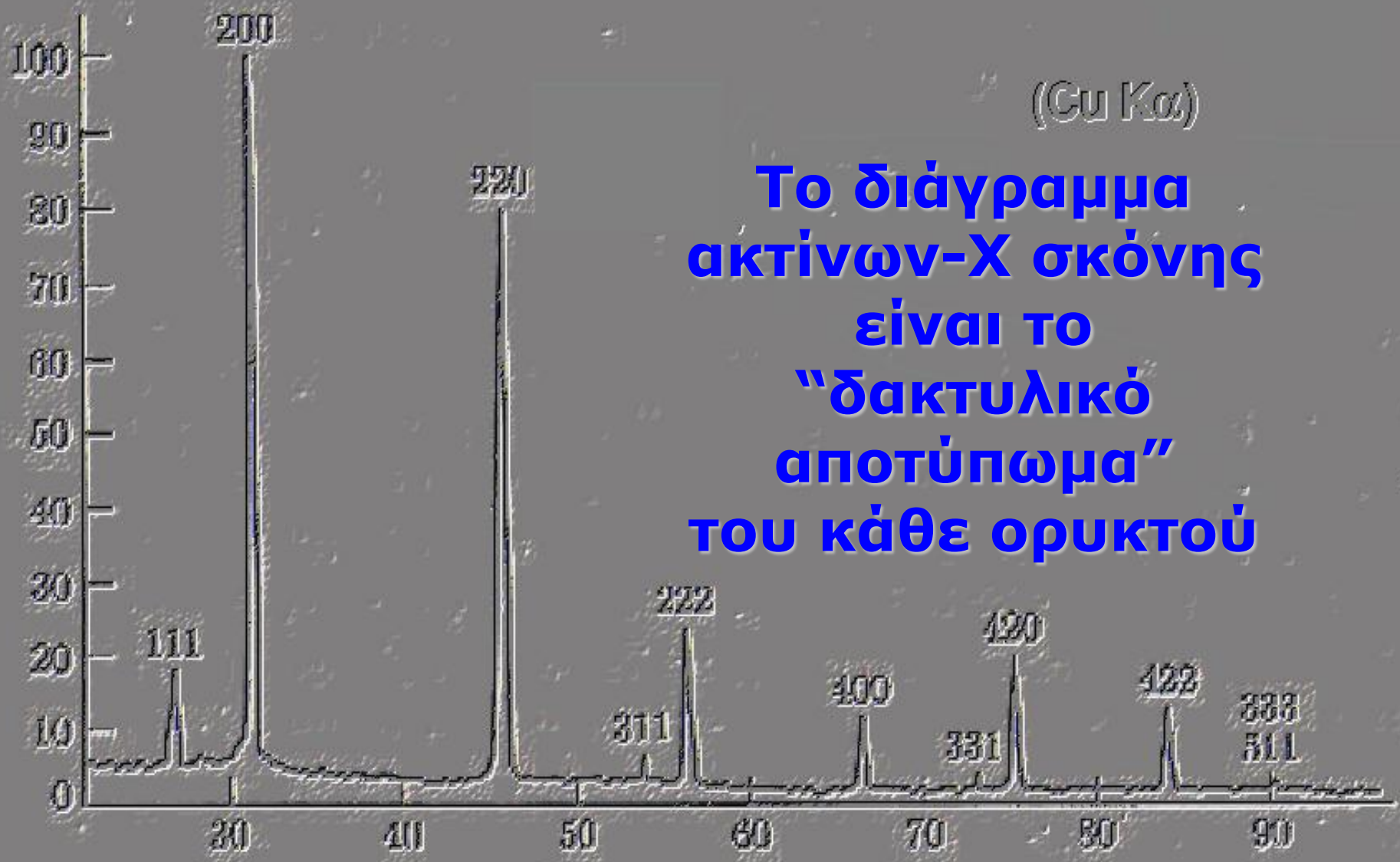
$$\lambda (\text{CuK}_\alpha) = 1.54 \text{ \AA},$$

$$a_0 (\text{NaCl}) = 5.6402 \text{ \AA}$$

$$2\theta_{220} = 45.45^\circ$$

$$d_{220} = 1.994$$





(Cu K α)

**Το διάγραμμα
ακτίνων-Χ σκόνης
είναι το
“δακτυλικό
αποτύπωμα”
του κάθε ορυκτού**

JCPDS Card

Quality of data

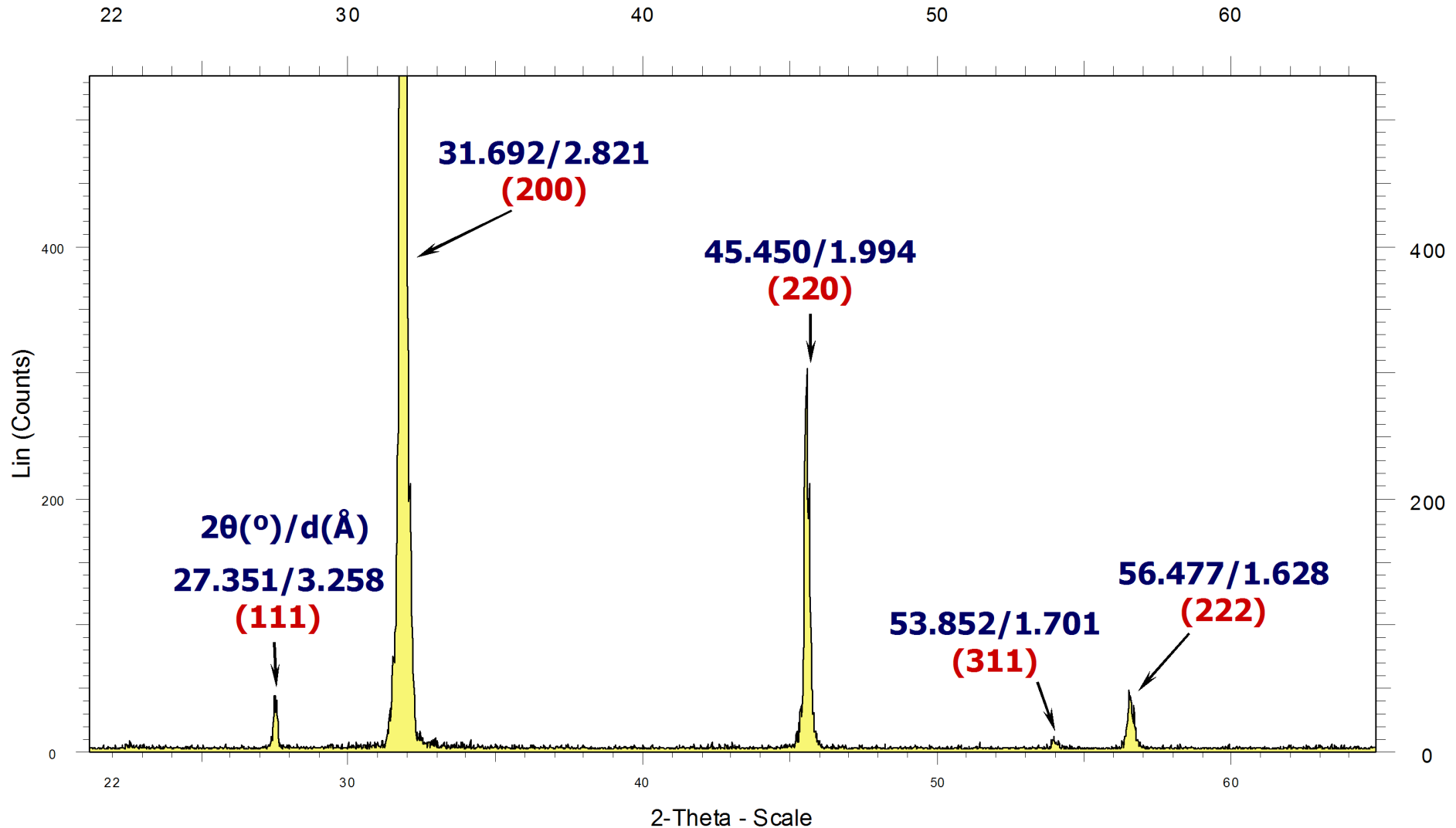
NaCl

d	2.82	1.99	1.63	3.26	NaCl			★		
I/I_1	100	55	15	13	Sodium Chloride			(Halite)		
Rad. CuK α_1 λ 1.5405 Filter Ni Dia.					d A	I/I_1	hkl	d A	I/I_1	hkl
Cut off I/I_1 Diffractometer I/I cor.					3.258	13	111			
Ref. Swanson and Fuyat, NBS Circular 539, Vol. 2, 41 (1953)					2.821	100	200			
Sys. Cubic S.G. Fm3m (225)					1.994	55	220			
					1.701	2	311			
a ₀ 5.6402 b ₀ c ₀ A C					1.628	15	222			
α β γ Z 4 Dx 2.164					1.410	6	400			
Ref. Ibid.					1.294	1	331			
eα nωβ 1.542 ey Si gn					1.261	11	420			
					1.1515	7	422			
2ν D mp Color Colorless					1.0855	1	511			
Ref. Ibid					0.9969	2	440			
An ACS reagent grade sample recrystallized twice from hydrochloric acid.					0.9533	1	531			
					0.9401	3	600			
					0.8917	4	620			
					0.8601	1	533			
					0.8503	3	622			
X-ray pattern at 26°C					0.8141	2	444			
Merck Index, 8th Ed., p. 956										
Halite-galena-periclase group.										

1.file number 2.three strongest lines 3.lowest-angle line 4.chemical formula and name 5.data on diffraction method used 6.crystallographic data 7.optical and other data 8.data on specimen 9.data on diffraction pattern.

*Joint Committee on Powder Diffraction Standards, JCPDS (1969)
Replaced by International Centre for Diffraction Data, ICDF (1978)*

NaCl



NaCl - File: NaCl.raw - Type: 2Th/Th locked - Start: 3.000 ° - End: 65.000 ° - Step: 0.020 ° - Step time: 1. s - Temp.: 25 °C (Room) - Time Started: 14 s - 2-Theta: 3.000 ° - Theta: 1.500 ° - Chi: 0.0

NaCl: $a_0 = 5.6402 \text{ \AA}$

$$(n)\lambda = 2 * d * \sin\theta$$

$$\lambda = 1.54056 \text{ \AA}$$

$$\theta = 2\theta/2$$

$$\sin 2\theta = \sin(\theta * \pi / 180)^2$$

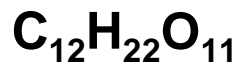
$$d = \lambda / (2 * \sin(\theta * \arccos(-1) / 180))$$

	θ	2θ	Cps	$\sin 2\theta$	I rel	d	$1 * \sin 2\theta / \sin 2\theta_{\min}$	$2 * \sin 2\theta / \sin 2\theta_{\min}$	$h^2 + k^2 + l^2$	hkl	$\alpha(\text{\AA})$
1	13.676	27.351	50	0.0559	4	3.258	1.487	2.974	2	110	4.6076
2	15.846	31.692	1280	0.075	100	2.821	1.984	3.967	4	200	5.6420
3	22.725	45.450	420	0.149	33	1.994	3.970	7.941	8	220	5.6397
4	26.926	53.852	10	0.205	1	1.701	5.456	10.911	11	311	5.6416
5	28.239	56.477	40	0.224	3	1.628	5.956	11.912	12	222	5.6396

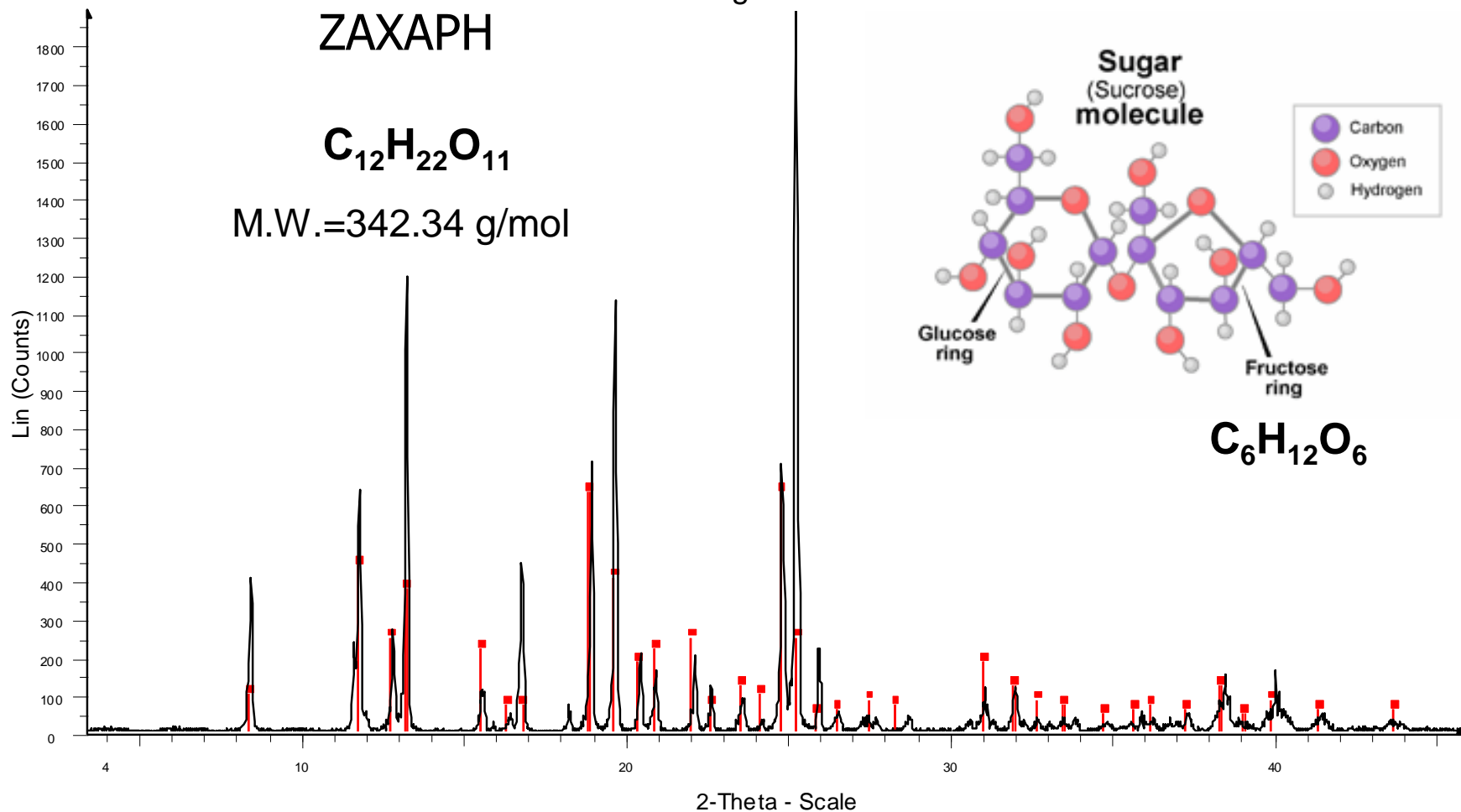
$$\text{NaCl: } a_0 = 5.6402 \text{ \AA}$$

Sugar

ZAXAPH

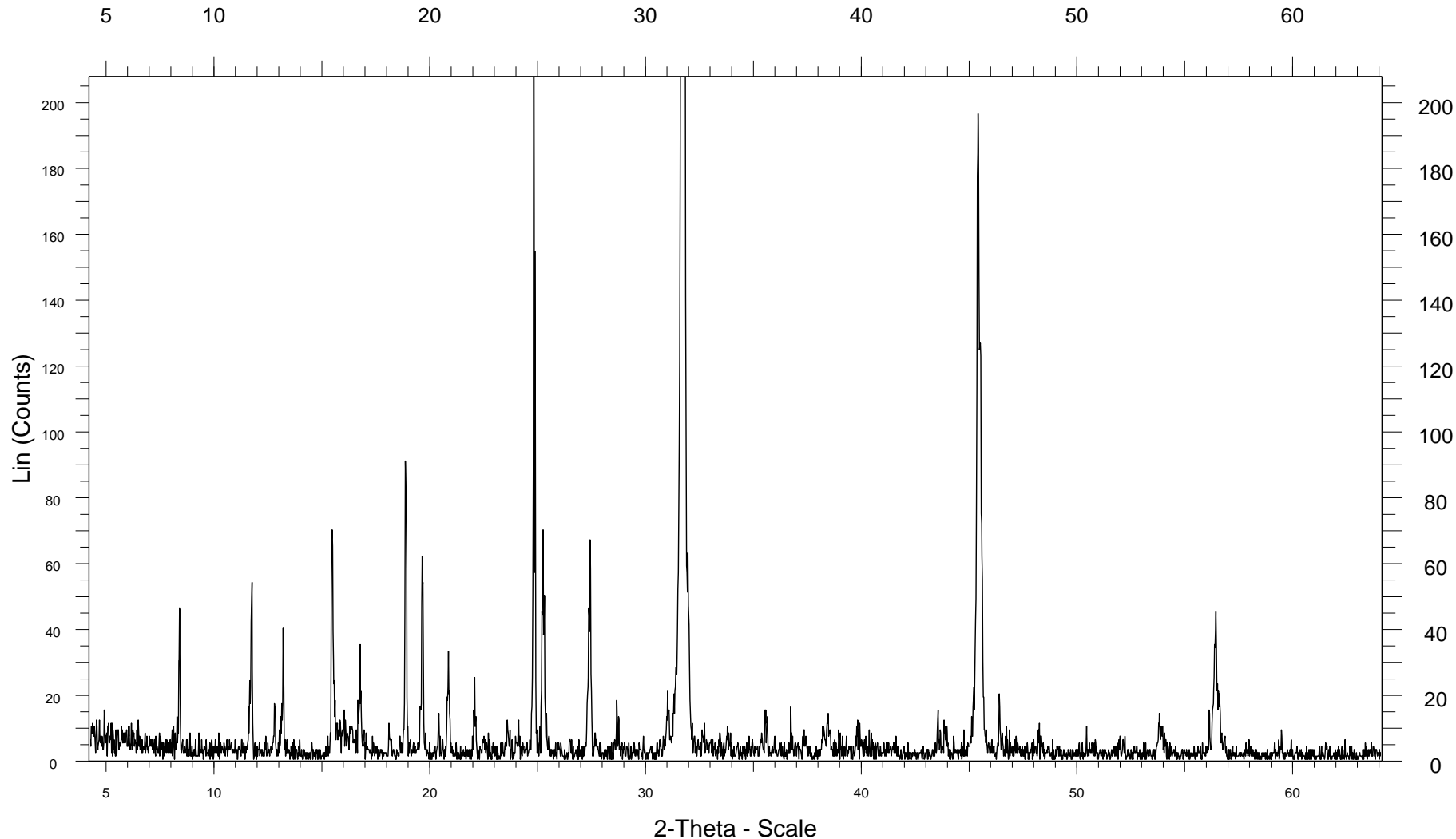


M.W.=342.34 g/mol



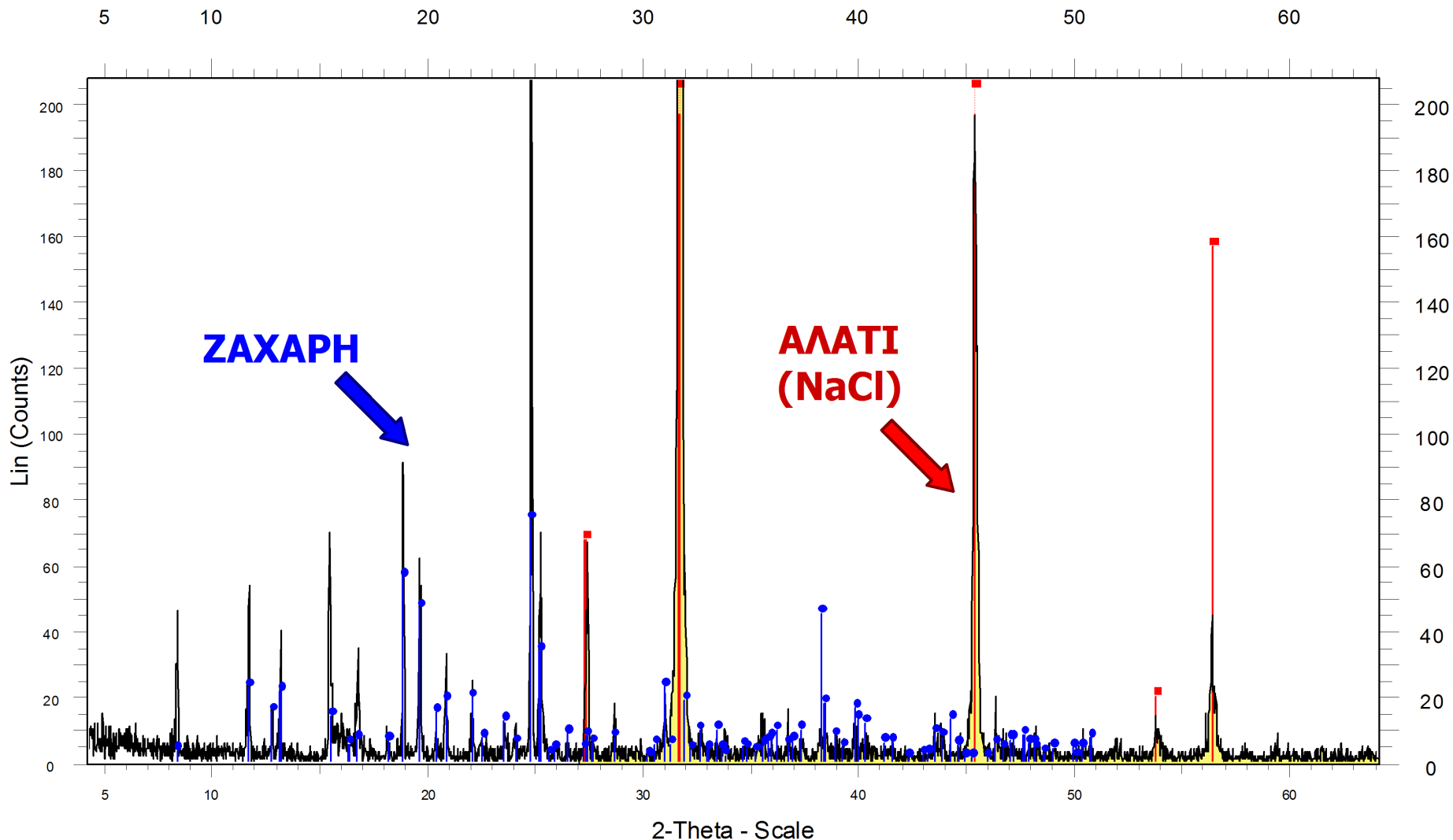
W Sugar - File: Sugar.raw - Type: 2Th/Th locked - Start: 3.000° - End: 65.000° - Step: 0.020° - Step time: 1. s - Temp.: 25 °C(Room) - Time Started: 15 s - 2-Theta: 3.000° - Theta: 1.500° - Chi: 0.00° - Phi: 0.0
Operations: Import
00-006-0142 (D) - Sucrose - C12H22O11 - Y: 33.38 % - d x by: 1. - WL: 1.5406 - 0 -

NaCl: 50% + Sugar: 50%



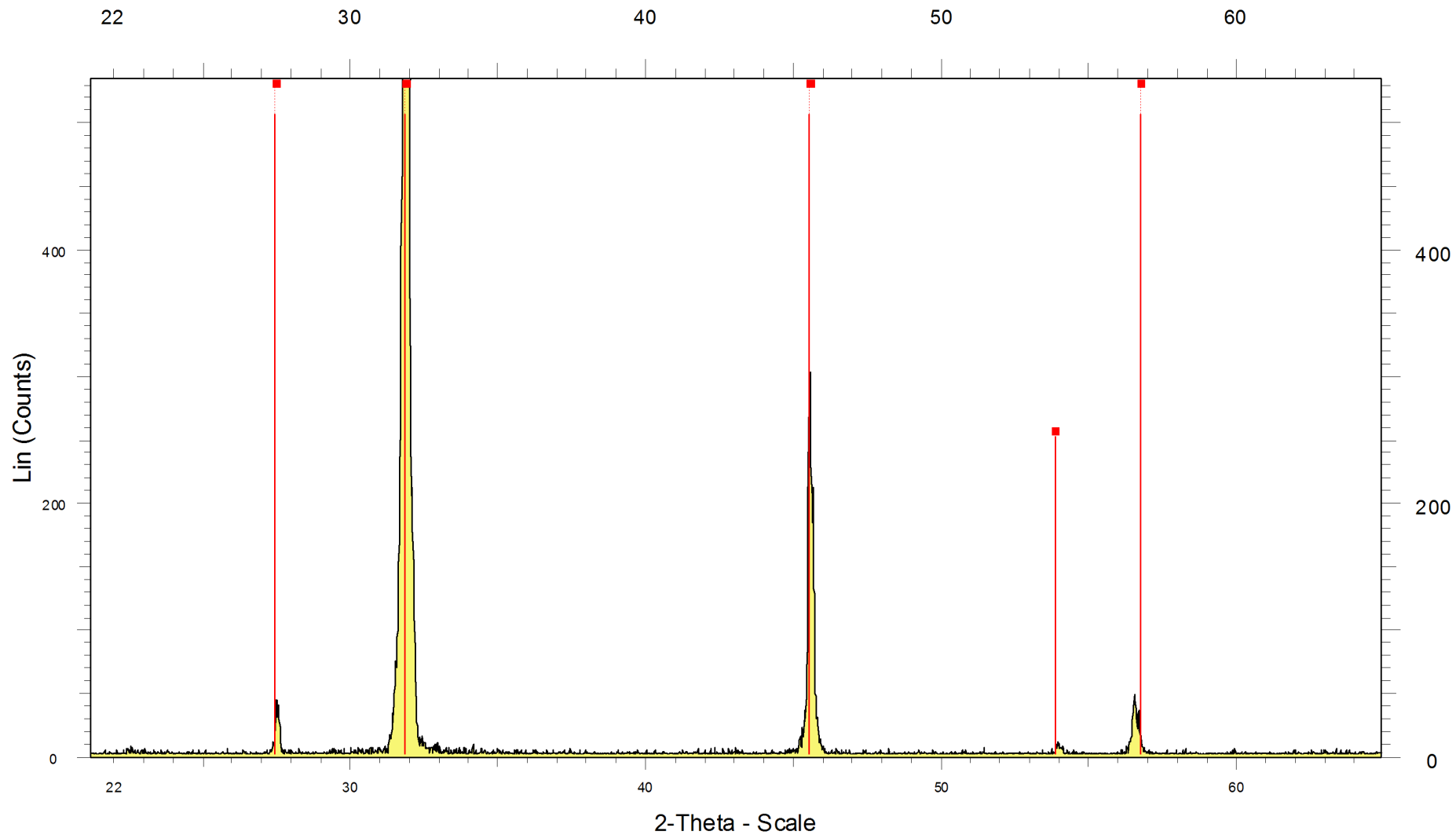
NaCl+Sugar - File: NaCl+Sugar.raw - Type: 2Th/Th locked - Start: 3.000 ° - End: 65.000 ° - Step: 0.020 ° - Step time: 1. s - Temp.: 25 °C (Room) - Time Started: 14 s - 2-Theta: 3.000 ° - Theta: 1.

NaCl: 50% + Sugar: 50%

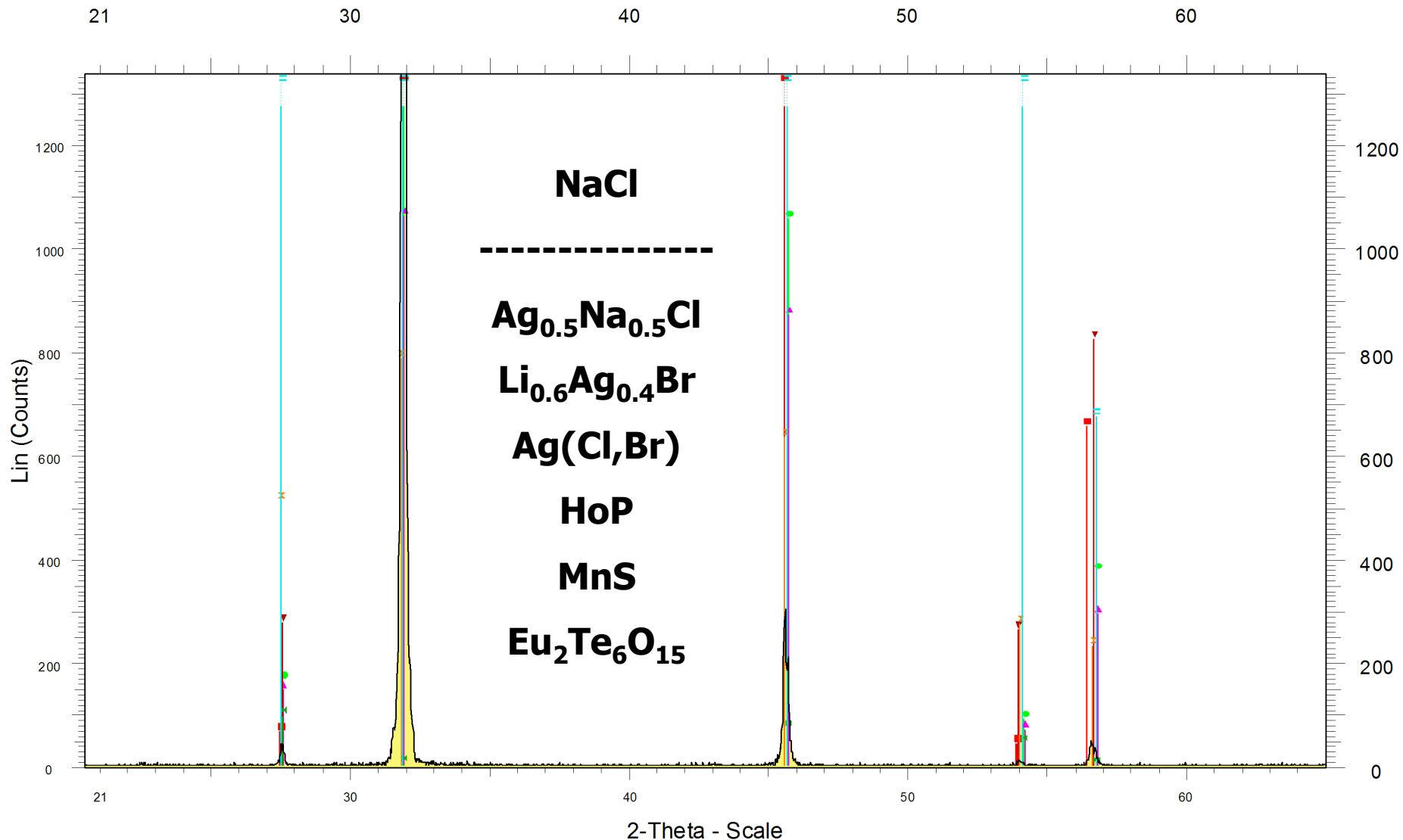


NaCl+Sugar - File: NaCl+Sugar.raw - Type: 2Th/Th locked - Start 3.000 ° - End: 65.000 ° - Step: 0.020 ° - Step time: 1. s - Temp.: 25 °C (Room) - Time Started: 14 s - 2-Theta: 3.000 ° - Theta: 1.
00-005-0628 (*) - Halite, syn - NaCl - Y: 30.57 % - d x by: 1. - WL: 1.5406 - 0 - I/Ic PDF 4.4 - S-Q 56.5 % -
00-024-1977 (*) - Sucrose - C12H22O11 - Y: 3.74 % - d x by: 1. - WL: 1.5406 - 0 - I/Ic PDF 0.7 - S-Q 43.5 % -

NaCl



NaCl - File: NaCl.raw - Type: 2Th/Th locked - Start: 3.000 ° - End: 65.000 ° - Step: 0.020 ° - Step time: 1. s - Temp.: 25 °C (Room) - Time Started: 14 s - 2-Theta: 3.000 ° - Theta: 1.500 ° - Chi: 0.0
00-002-0818 (D) - Halite - NaCl - Y: 36.38 % - d x by: 1. - WL: 1.5406 - 0 - I/Ic PDF 1. - S-Q 100.0 % -



NaCl - File: NaCl.raw - Type: 2Th/Th locked - Start: 3.000 ° - End: 65.000 ° - Step: 0.020 ° - Step time: 1. s - Temp.: 25 °C (Room) - Time Started: 14 s - 2-Theta: 3.000 ° - Theta: 1.500 ° - Chi: 0.0

00-022-0898 (*) - Silver Sodium Chloride - Ag_{0.5}Na_{0.5}Cl - Y: 18.17 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 3.3 - S-Q 4.0 % -

00-026-0856 (C) - Lithium Silver Bromide - Li_{0.6}Ag_{0.4}Br - Y: 14.21 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 1. - S-Q 10.4 % -

00-014-0255 (N) - Chlorargyrite, bromian - Ag(Cl,Br) - Y: 21.21 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 1. - S-Q 15.6 % -

00-001-0994 (D) - Halite - NaCl - Y: 43.10 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 1. - S-Q 31.7 % -

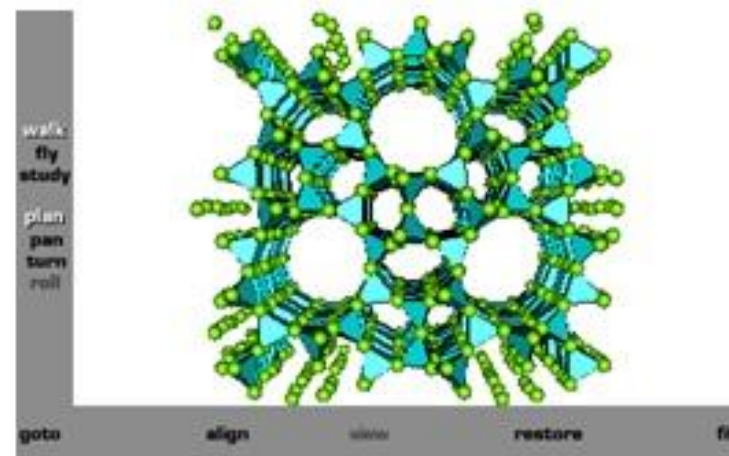
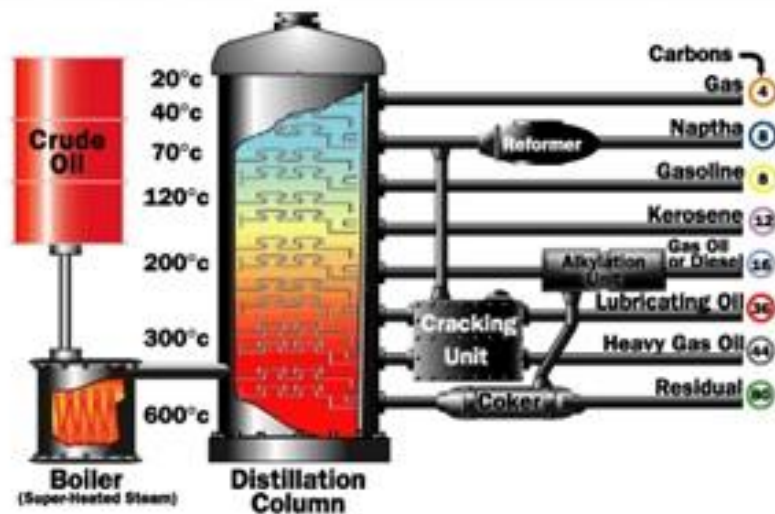
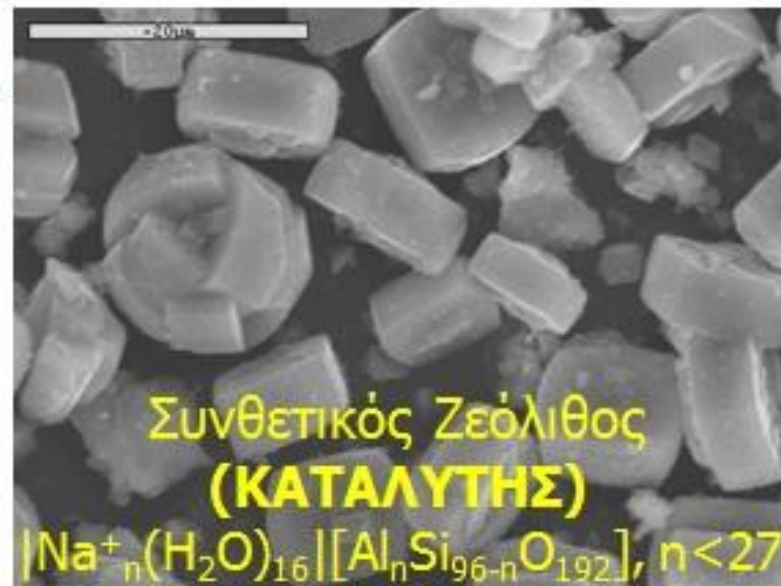
00-030-0616 (C) - Holmium Phosphide - HoP - Y: 10.54 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 19.6 - S-Q 0.4 % -

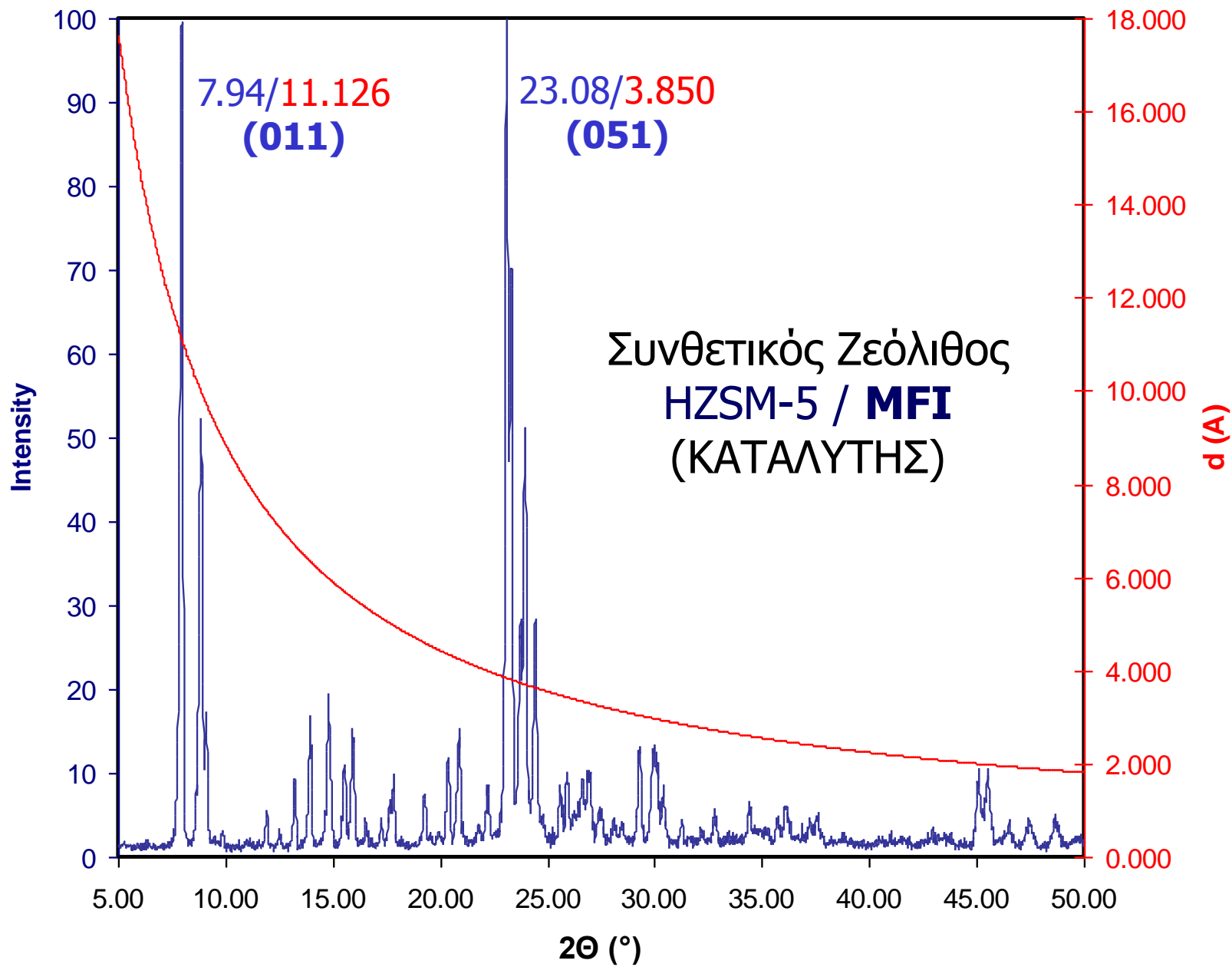
00-040-1288 (*) - Manganese Sulfide - MnS - Y: 1.29 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 1. - S-Q 0.9 % -

00-037-1399 (I) - Europium Tellurate - Eu₂Te₆O₁₅ - Y: 50.14 % - d x by: 1. - WL: 1.5406 - Cubic - I/Ic PDF 1. - S-Q 36.9 % -

ΚΑΤΑΛΥΤΙΚΗ ΔΙΑΣΠΑΣΗ ΠΕΤΡΕΛΑΙΟΥ

Παραγωγή κλασμάτων υδρογονανθράκων με κοντύτερες ανθρακικές αλυσίδες (π.χ. C8)

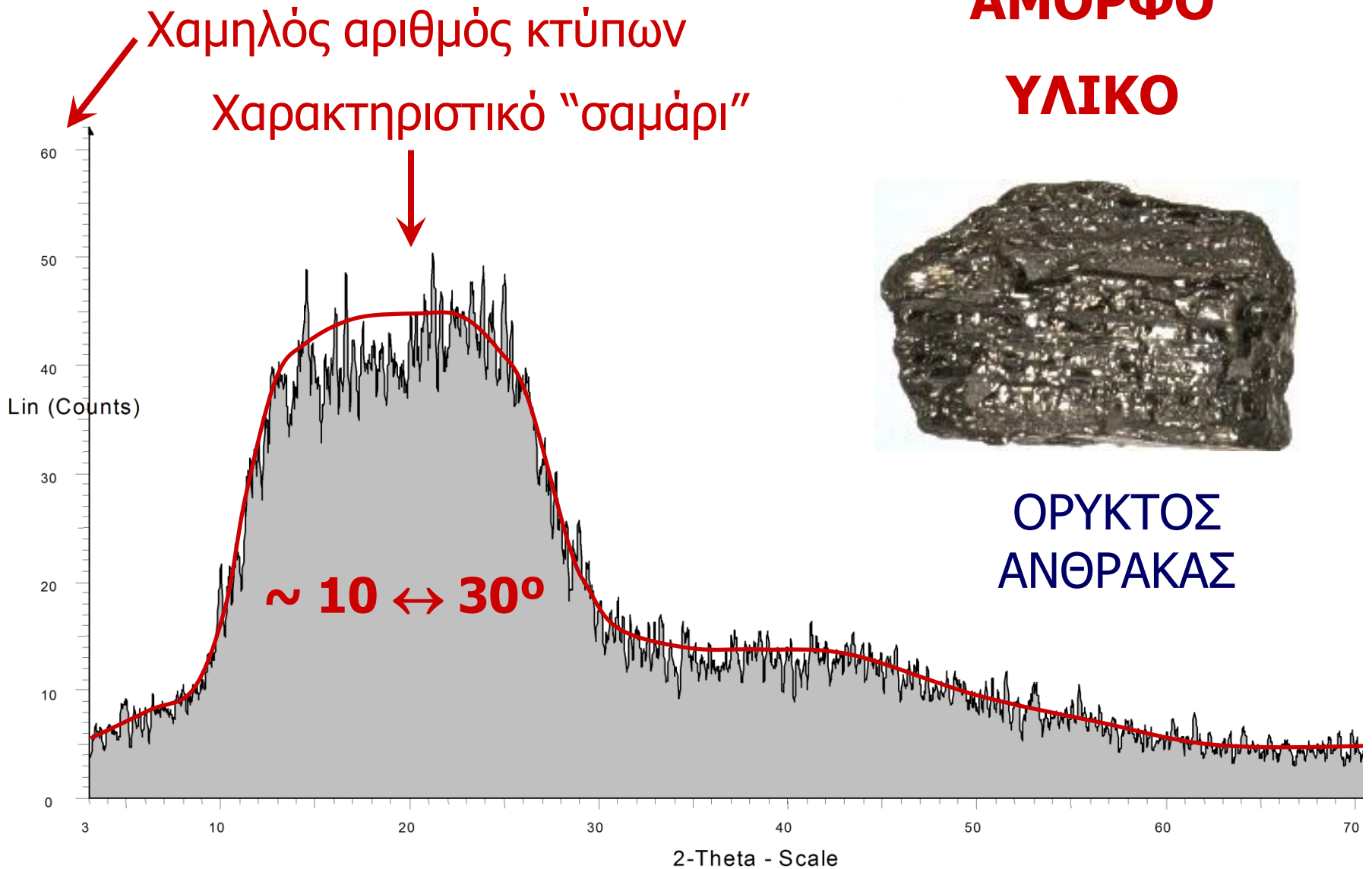




ΑΜΟΡΦΟ ΥΛΙΚΟ



ΟΡΥΚΤΟΣ
ΑΝΘΡΑΚΑΣ

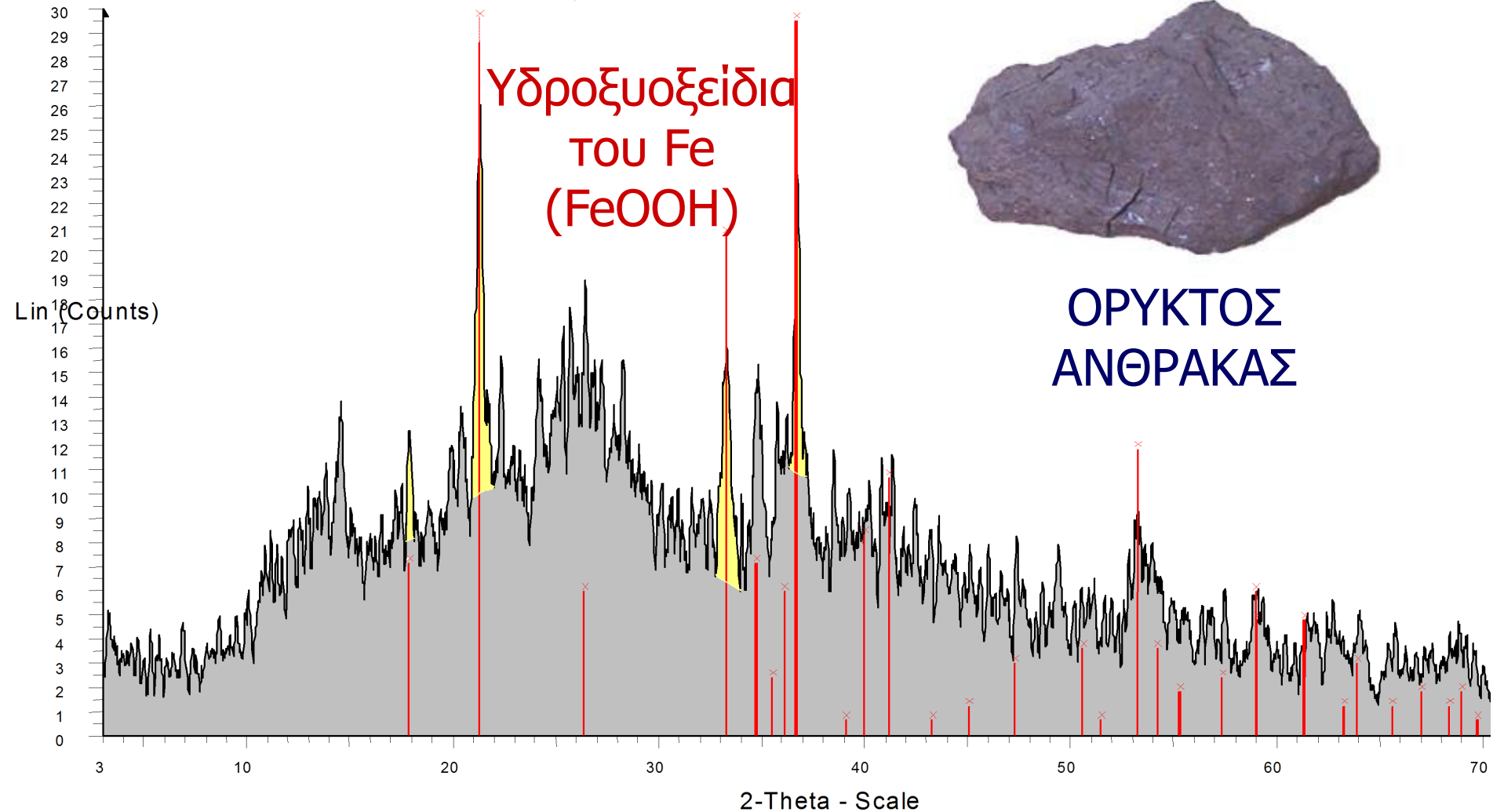


ΑΜΟΡΦΟ ΥΛΙΚΟ ΜΕ ΠΡΟΣΜΕΙΞΗ ΚΡΥΣΤΑΛΛΙΚΟΥ ΥΛΙΚΟΥ



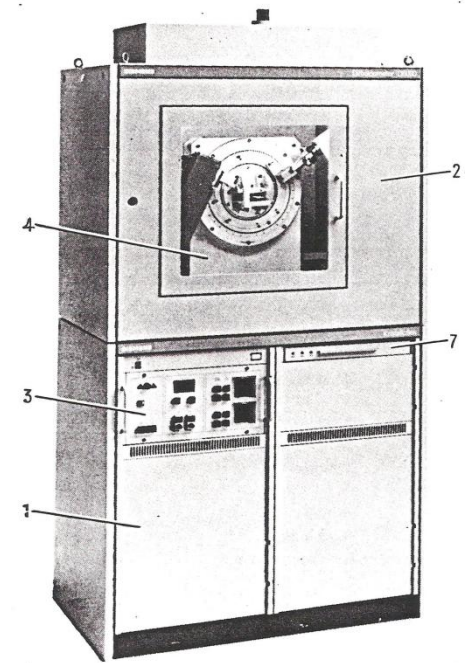
ΟΡΥΚΤΟΣ
ΑΝΘΡΑΚΑΣ

Υδροξυοξείδια
του Fe
(FeOOH)

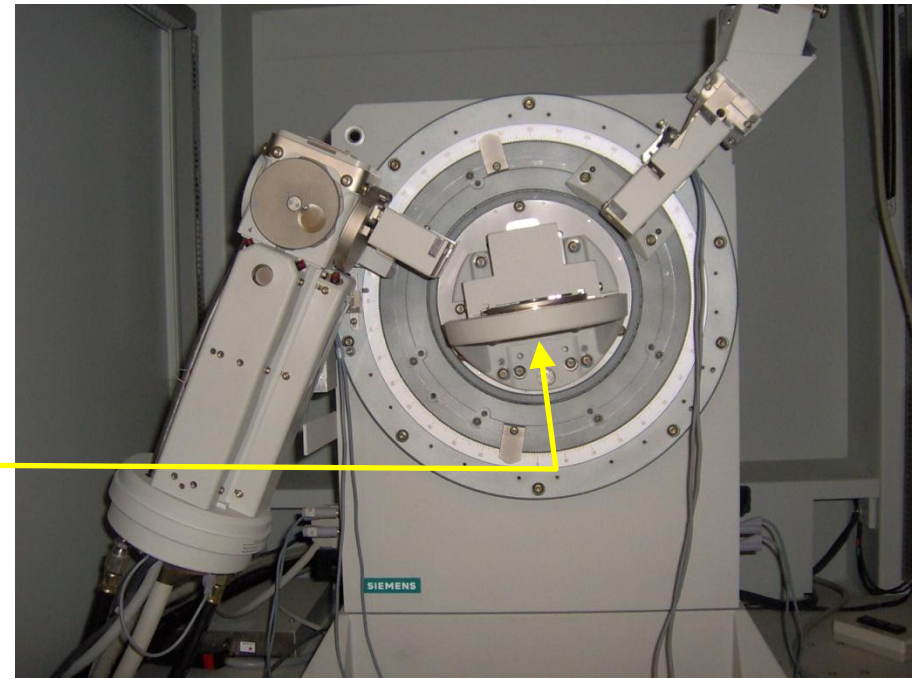
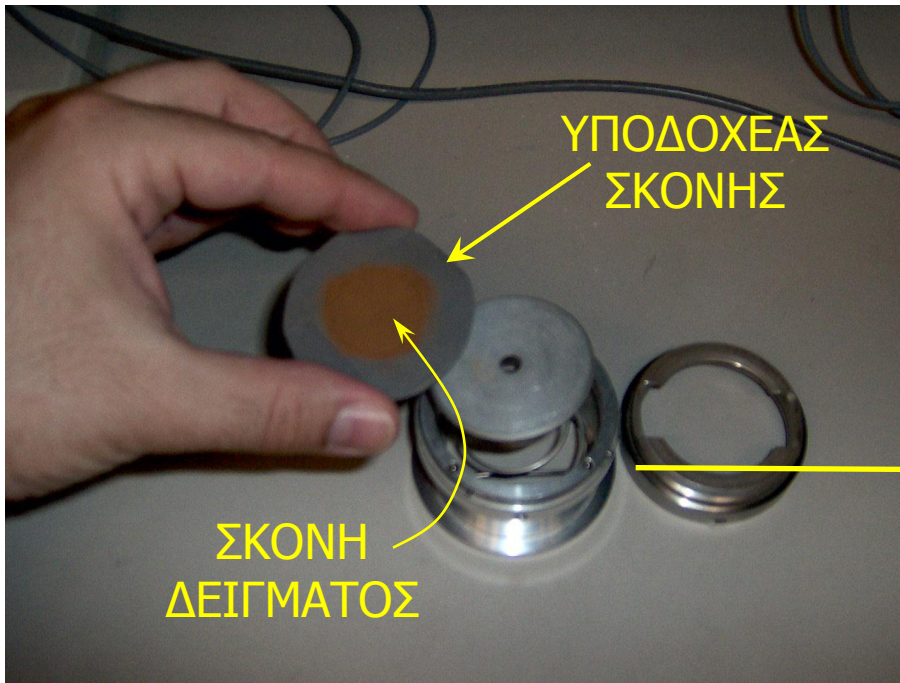


ΠΕΡΙΘΛΑΣΗ ΑΚΤΙΝΩΝ-Χ

Περιθλασίμετρο Ακτίνων-Χ Σκόνης

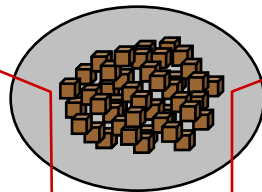


- | | | | |
|---|----------------------------------------------------------------------------------------------------------|---|---------------------------------------------------------------------------------------|
| 1 | Standgehäuse
Console
Console | 2 | Strahlenschutzgehäuse
Radiation protection
Boîtier de protection des radiations |
| 3 | Röntgeneratore KRISTALLOFLEX®
KRISTALLOFLEX® X-ray generator
Générateur de rayons X KRISTALLOFLEX® | | |
| 4 | Goniometer
Goniometer
Goniomètre | 7 | Bedienfeld
Control panel
Panneau de commande |



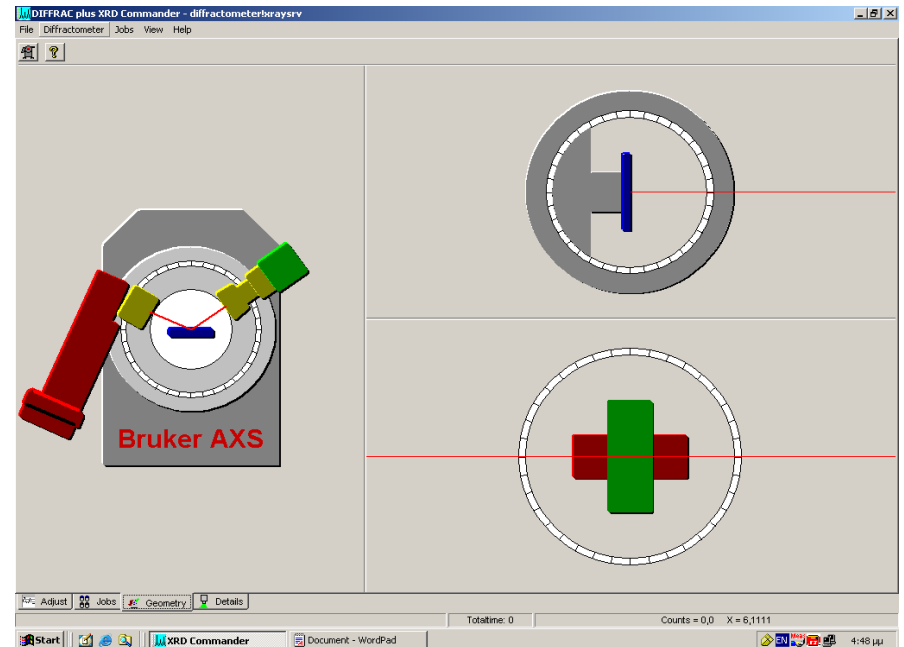
ΜΕΓΕΘΟΣ ΚΟΚΚΩΝ ΣΚΟΝΗΣ

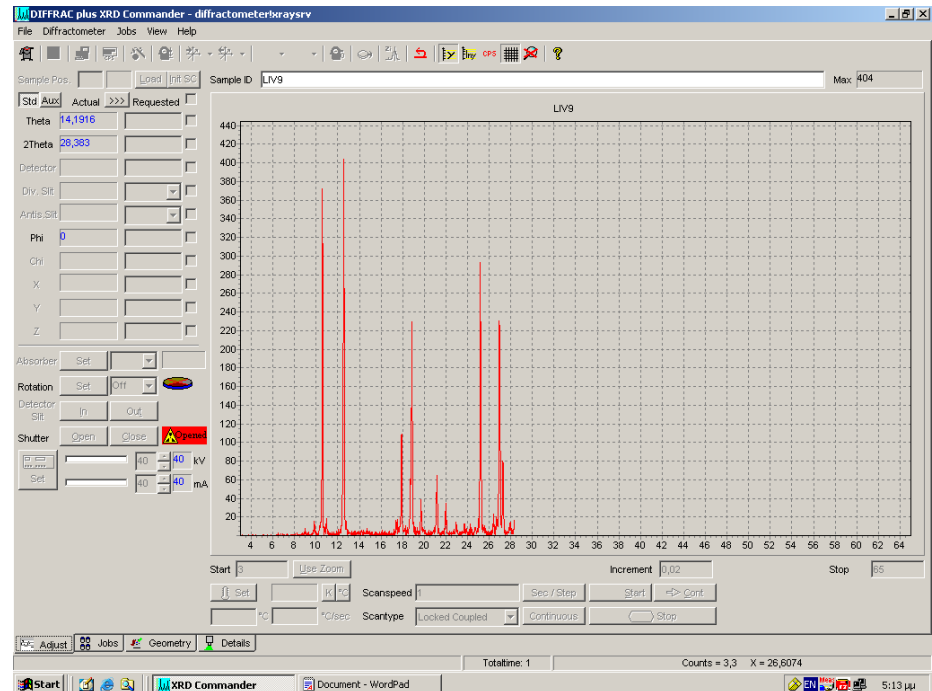
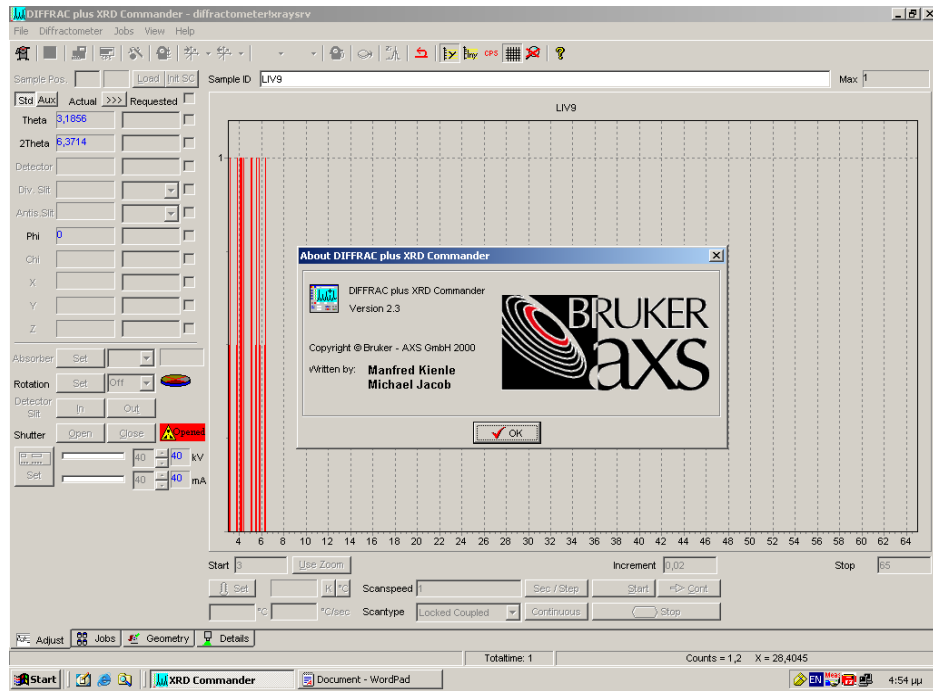
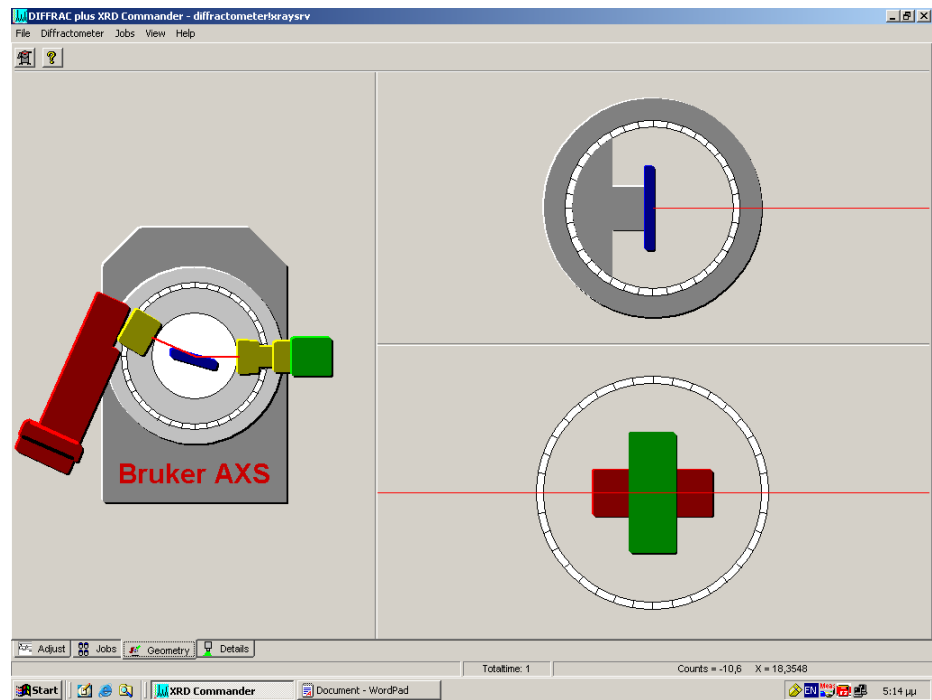
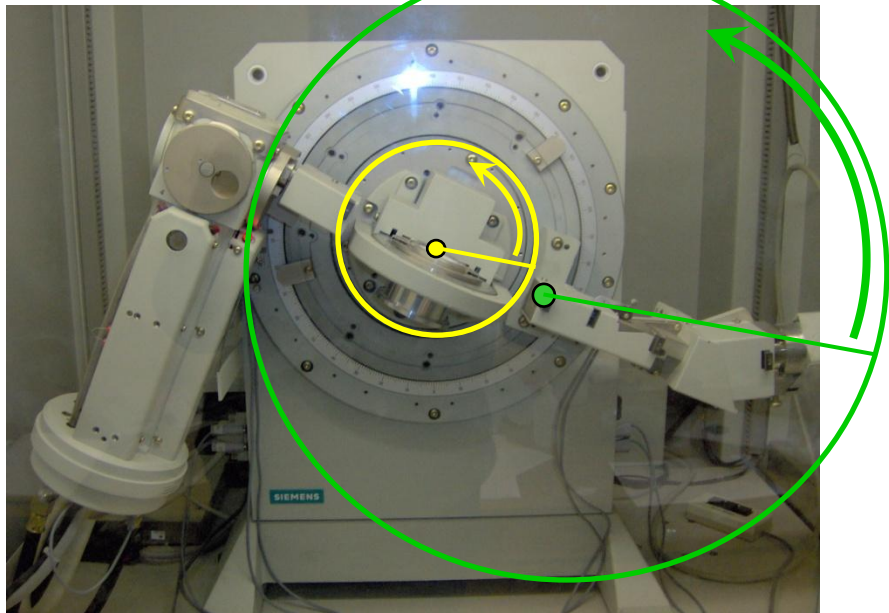
0.1 μm (100 nm) – 40 μm



ΔΙΕΥΡΥΝΣΗ
ΚΟΡΥΦΩΝ

ΜΕΙΩΜΕΝΗ
ΠΕΡΙΘΛΑΣΗ







Sample Pos. [] [] Load Infr SC Sample ID LIV9 Max 436

Std Aux Actual >>> Requested

Theta 32,505

2Theta 65,01

Detector

Div. Slit

Antis. Slit

Phi 0

Chi

X

Y

Z

Absorber Set

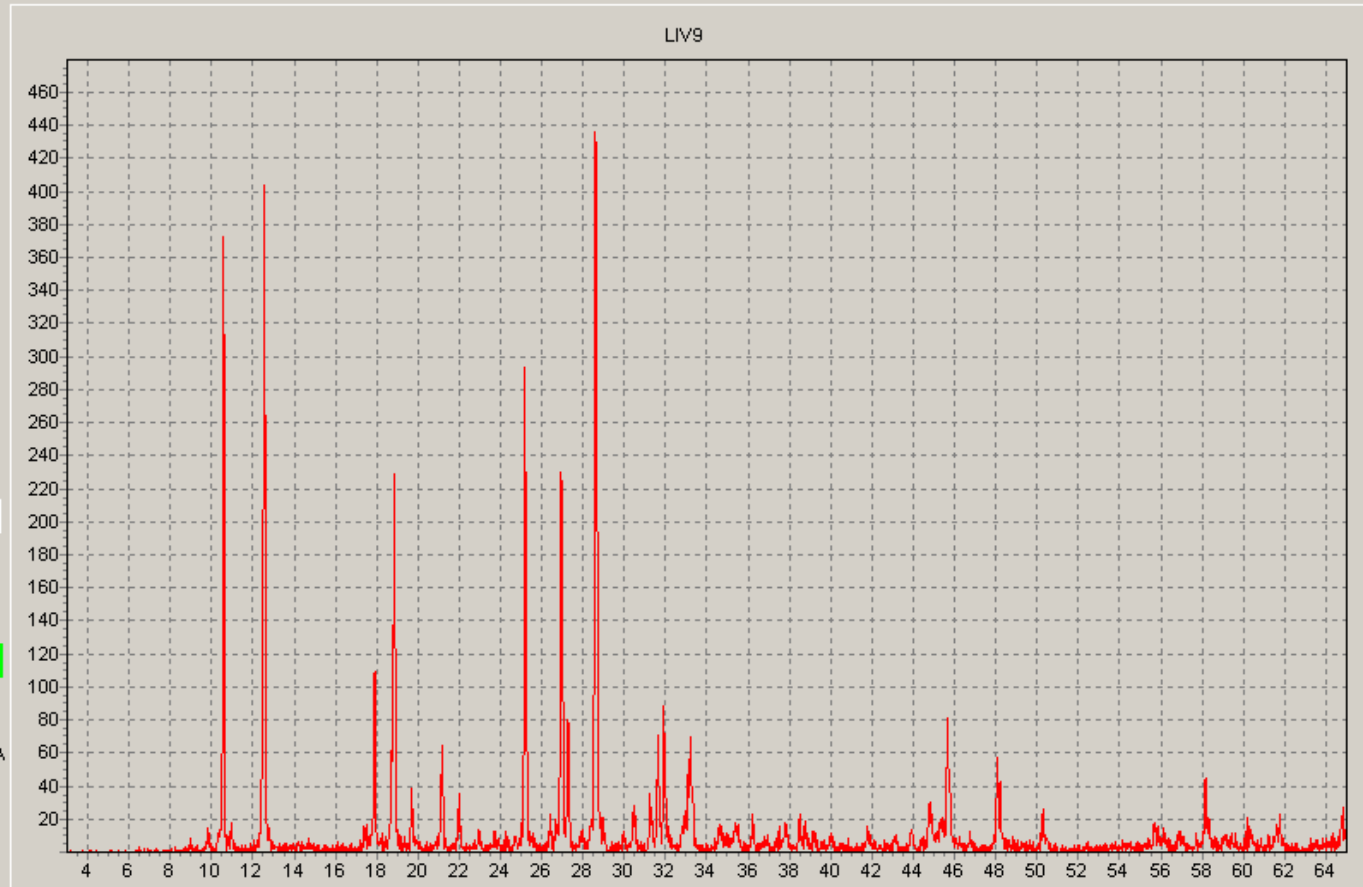
Rotation Set Off

Detector Slit In Out

Shutter Open Close

40 20 kV

40 5 mA



Start 3 Use Zoom Increment 0,02 Stop 65

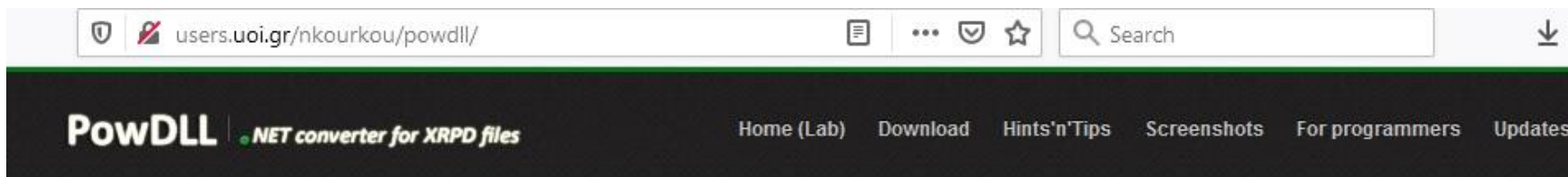
Set [] K °C Scanspeed 1 Sec / Step Start Cont

[] °C [] °C/sec Scantype Locked Coupled Continuous STOP Stop

Adjust Jobs Geometry Details

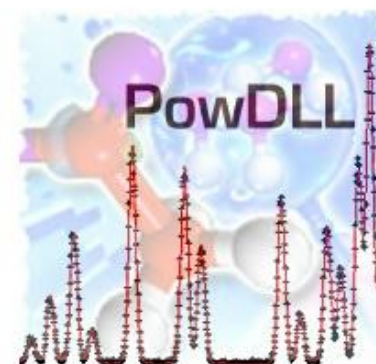
Totaltime: 1 Counts = 496,2 X = 50,4752

PowDLL



PowDLL is a .NET dynamic link library used for the interconversion procedure between variable formats of Powder X-Ray files. The DLL is capable of handling the most common file formats (binary and ASCII). The library can be used as a reusable component with any .NET language or as a **standalone utility**.

PowDLL can run on windows OS as long as they have dotnet runtimes version 2 (or later) installed (i.e., Windows Vista SP1 or later). Linux is also supported through Wine.



Imports

Bruker/Siemens RAW (versions 1,2,4), Bruker BRML, STOE RAW (plus multi-range files), Scintag RAW (plus multi-range files), Rigaku RAW, Shimadzu RAW, Philips RD, Philips SD, Scintag RD, Panalytical XRDML, INEL Binary,

Exports

Bruker/Siemens RAW (versions 1,2), Philips RD, Scintag ARD, Sietronics CPI, Riet7 DAT, DBWS, GSAS (CW STD), Jade MDI, Rigaku RIG, Philips UDF, UXD, XDA, XDD, Panalytical XRDML, ASCII XY Files, MS-Excel Multiple XY,

Citation

Please cite if you find PowDLL useful:

PowDLL, a reusable .NET component for interconverting powder diffraction data: Recent developments, N. Kourkoumelis, ICDD Annual Spring

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EXPO&more Workshop

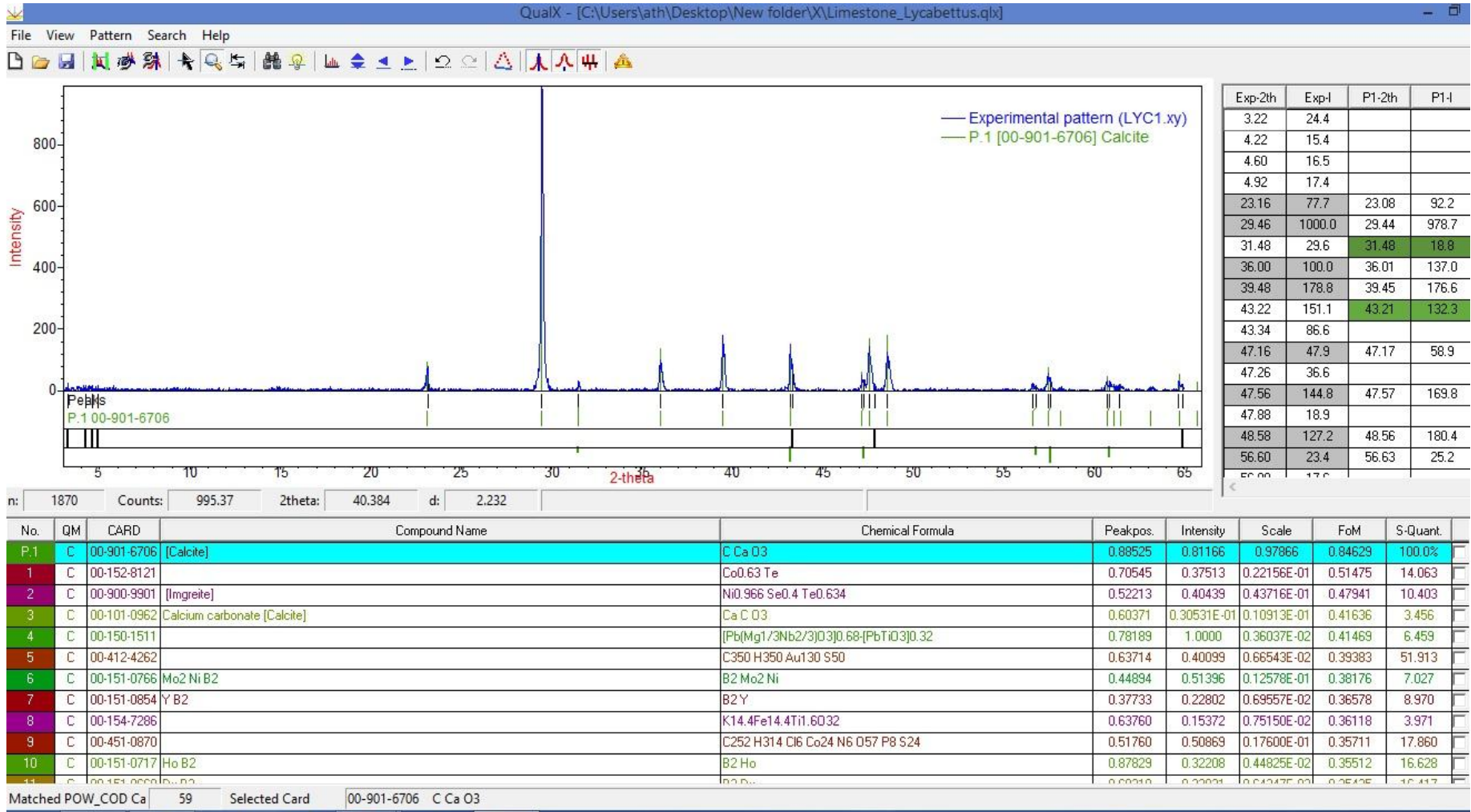
30 September – 3 October 2019

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QualX User Experience Survey

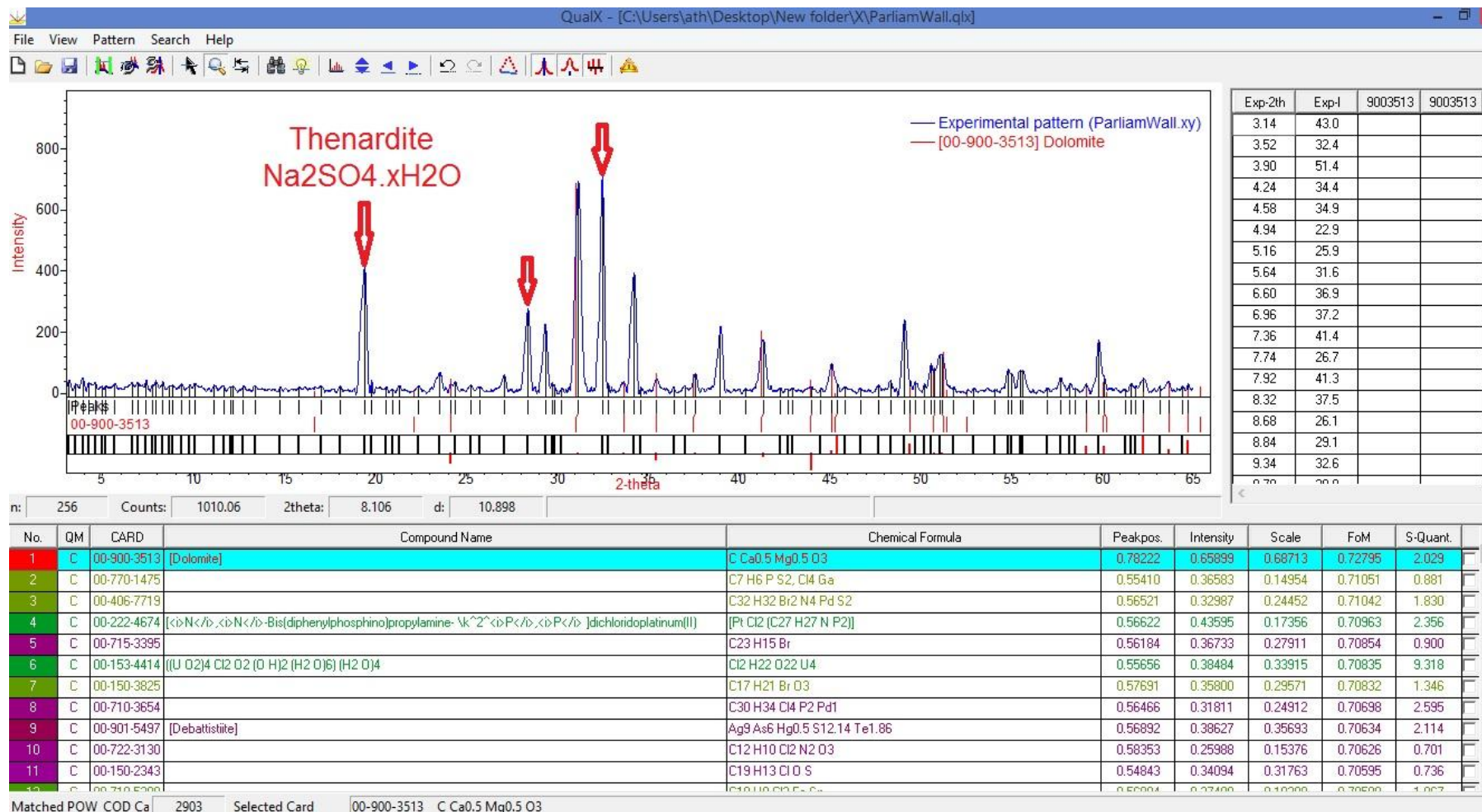


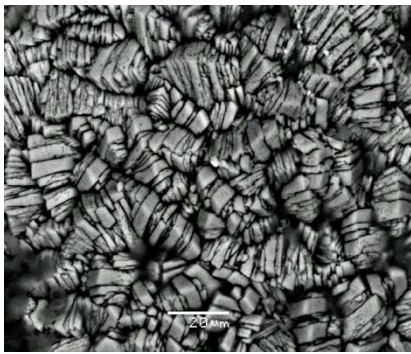
ΑΣΒΕΣΤΟΛΙΘΟΣ ΛΥΚΑΒΗΤΤΟΥ



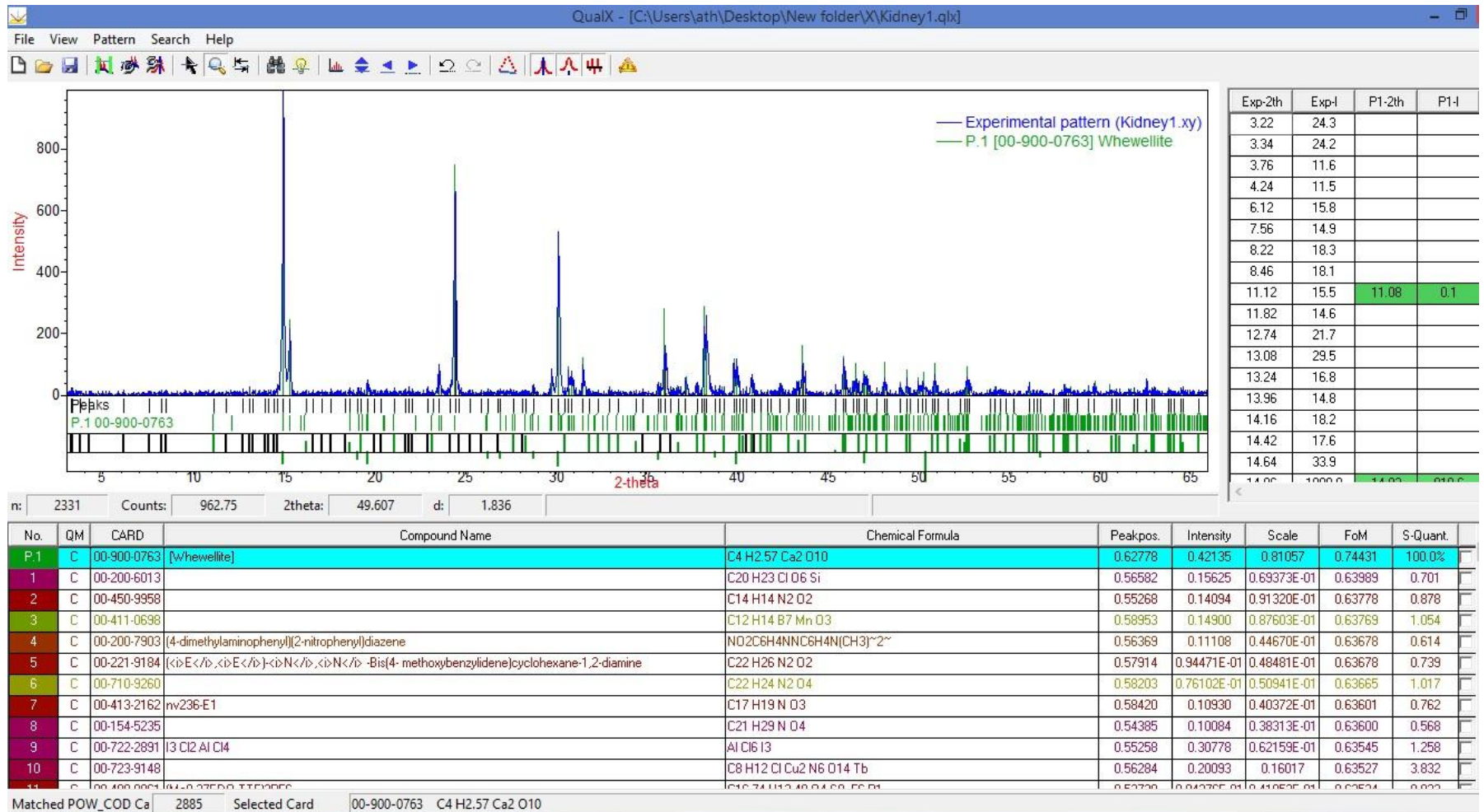


ΤΟΙΧΟΣ ΒΟΥΛΗΣ



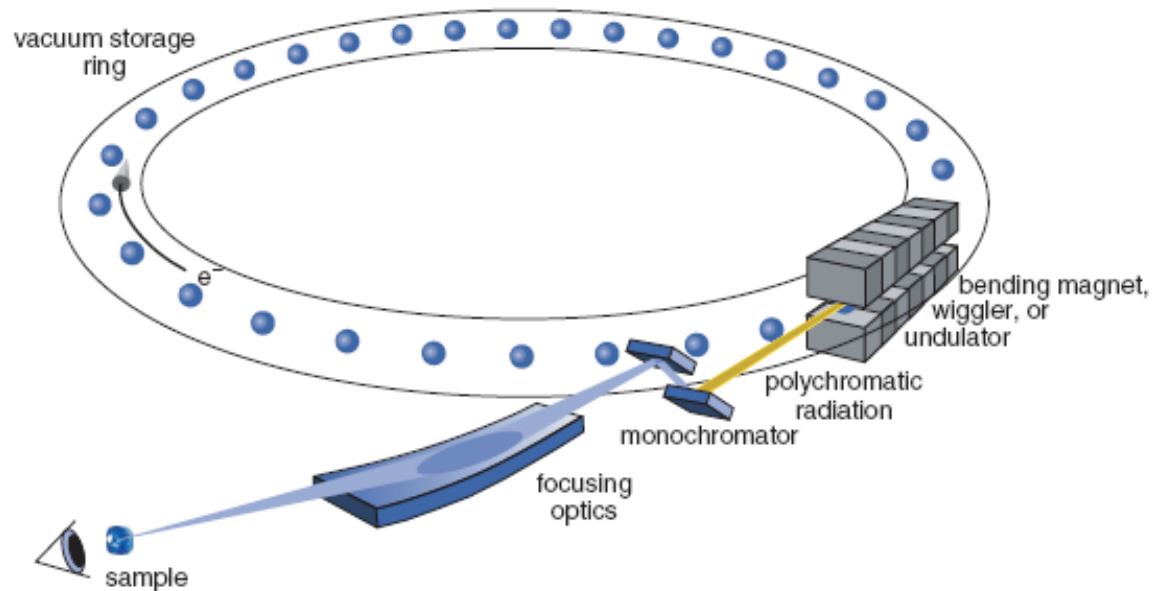


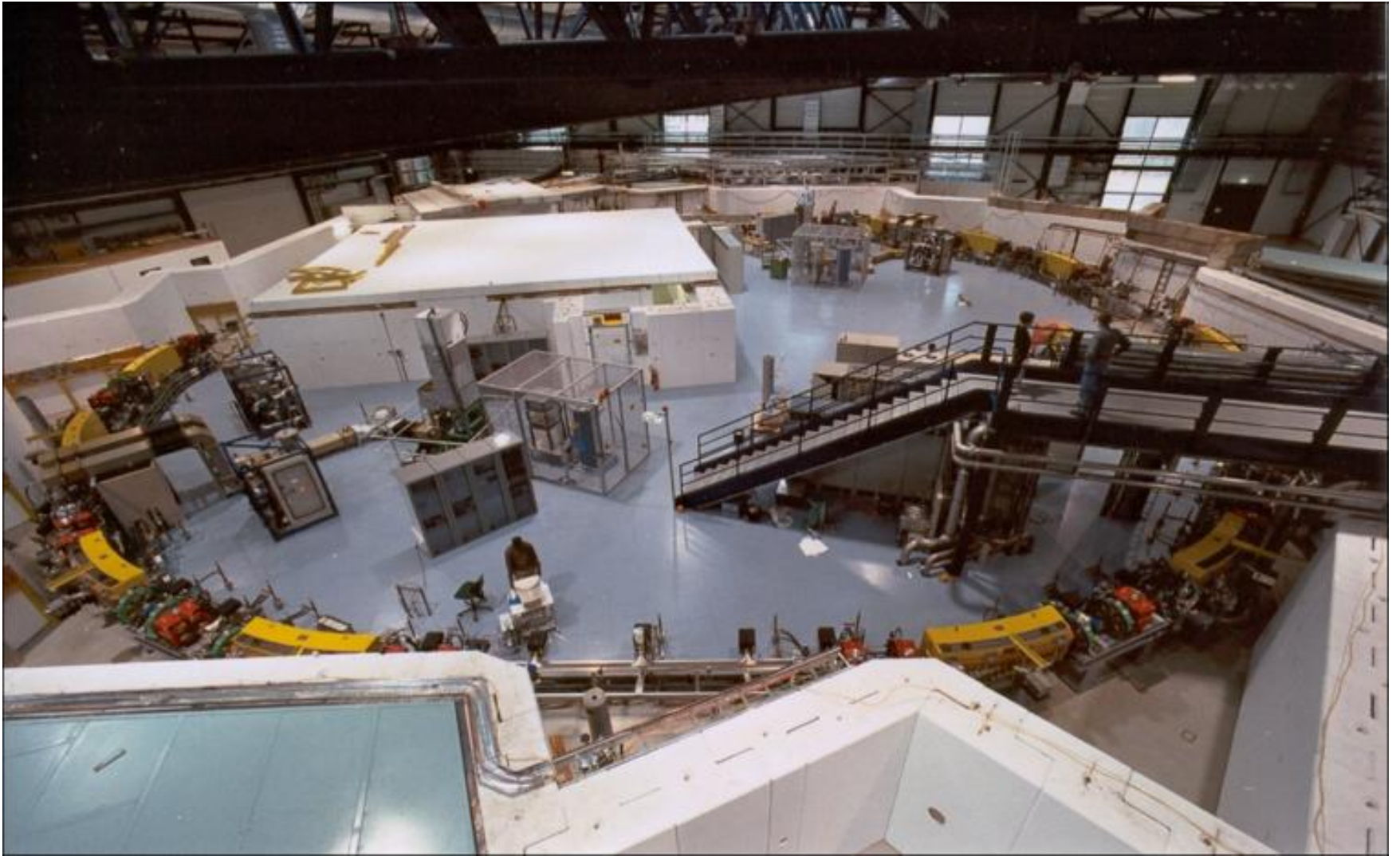
ΠΕΤΡΑ ΝΕΦΡΟΥ



ΠΕΡΙΘΛΑΣΗ ΑΚΤΙΝΩΝ-Χ ΜΕ ΧΡΗΣΗ ΑΚΤΙΝΟΒΟΛΙΑΣ ΣΥΝΧΡΟΤΡΟΝ (SR-XRD)

Ακτινοβολία που παράγεται από ηλεκτρόνια τα οποία κινούνται σε κλειστές τροχιές και με ταχύτητες που πλησιάζουν την ταχύτητα του φωτός ($0.985c$)



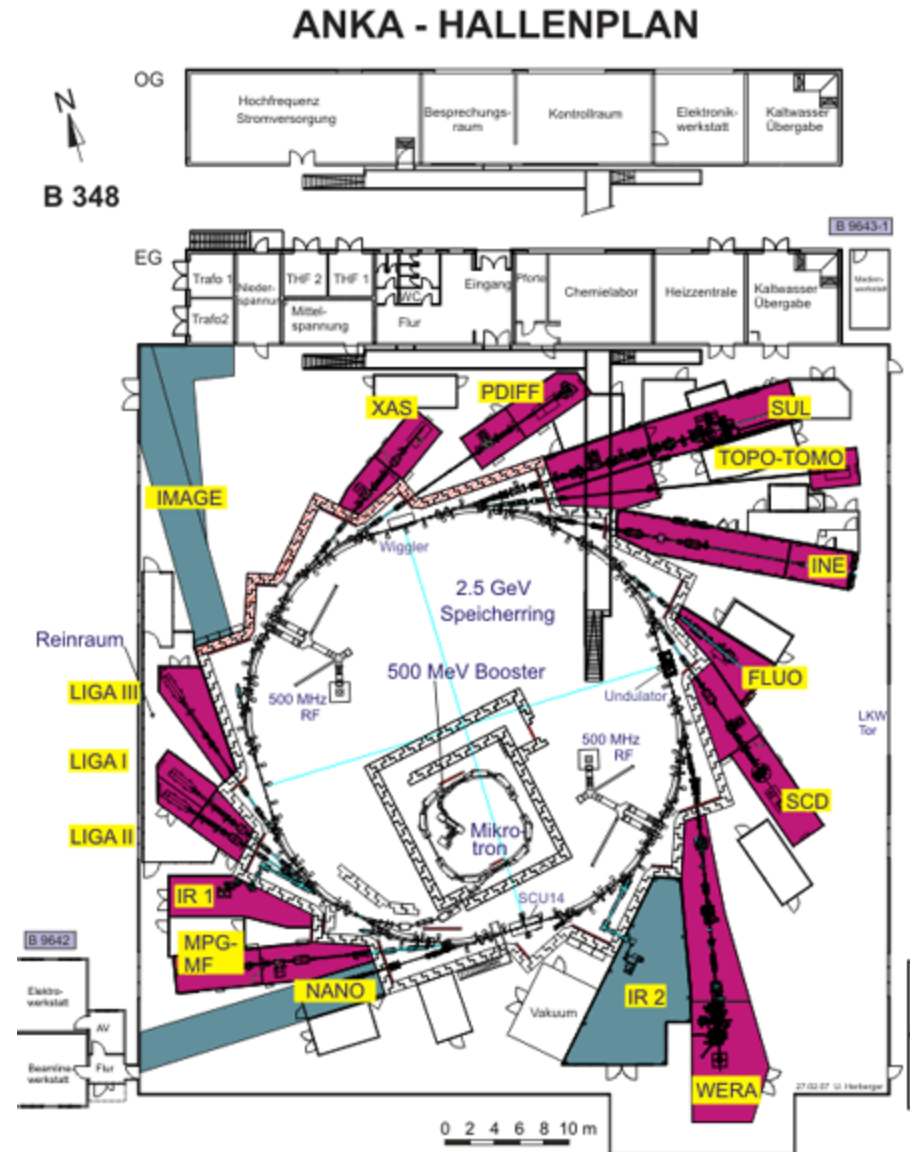


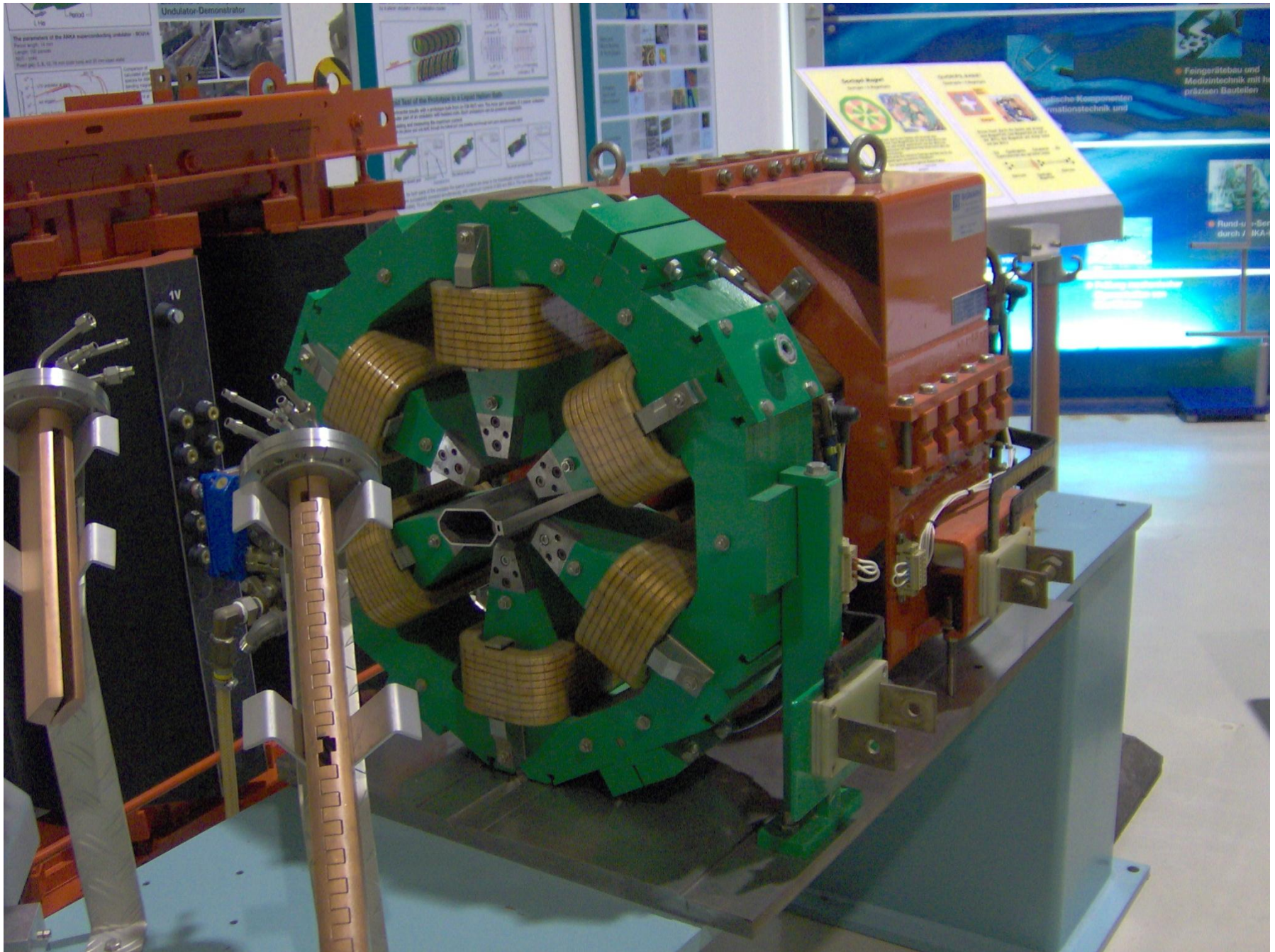
➤ Συνεχές φάσμα που εκτείνεται από το IR ως τις σκληρές ακτίνες-X

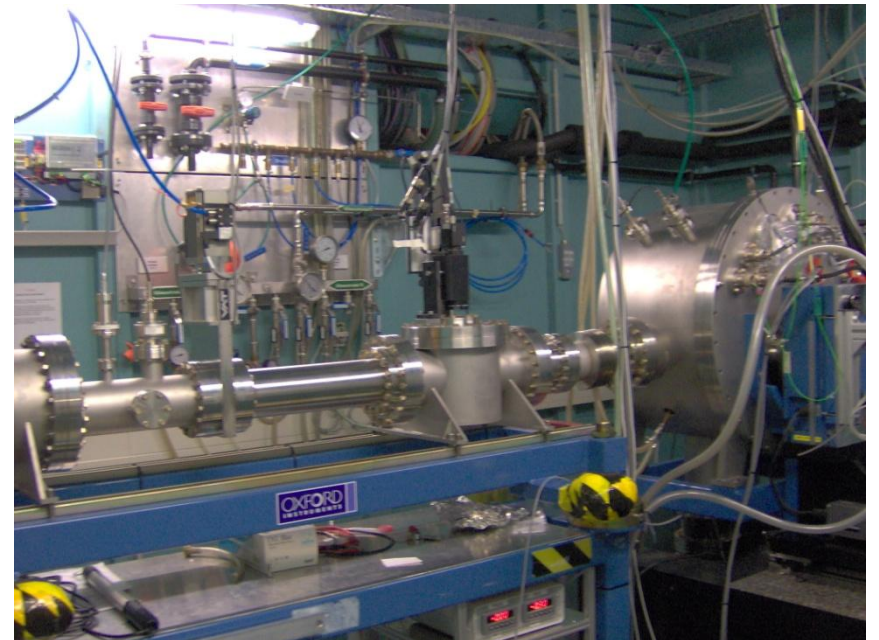
➤ Υψηλή ένταση (10^6 μεγαλύτερη από τις λυχνίες ακτίνων-X)

➤ Δέσμη πολύ μικρής διατομής και μικρής γωνιακής απόκλισης

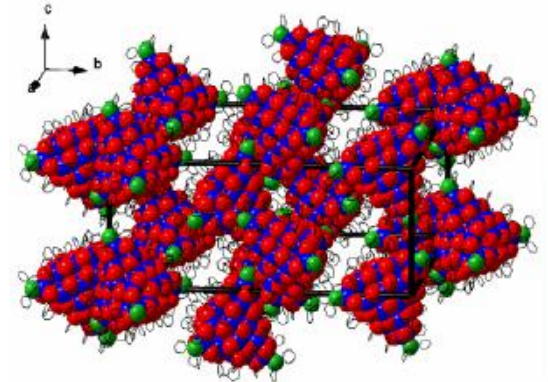
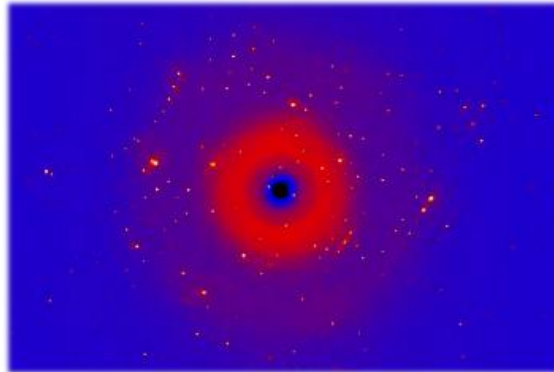
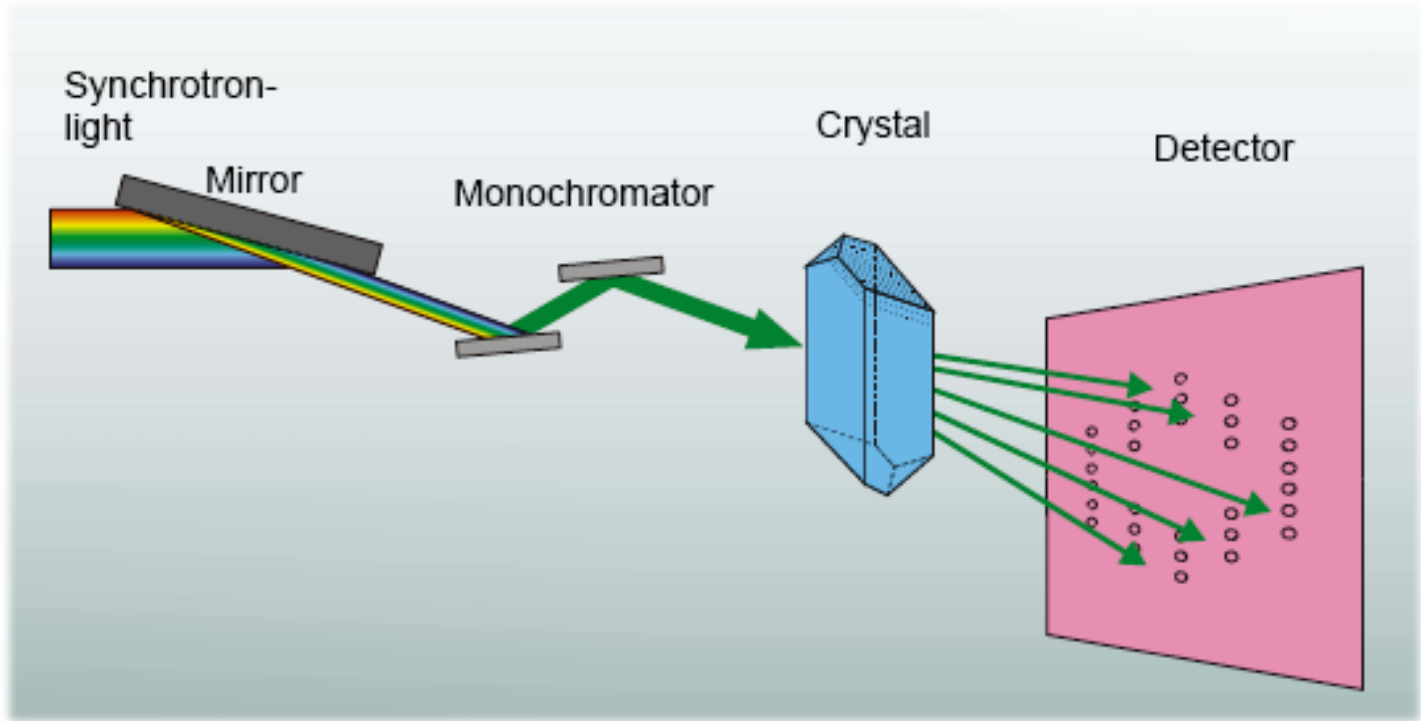
➤ Διαφορετικές γραμμές ακτινοβολίας (beamlines) για εκτέλεση ταυτόχρονων πειραμάτων



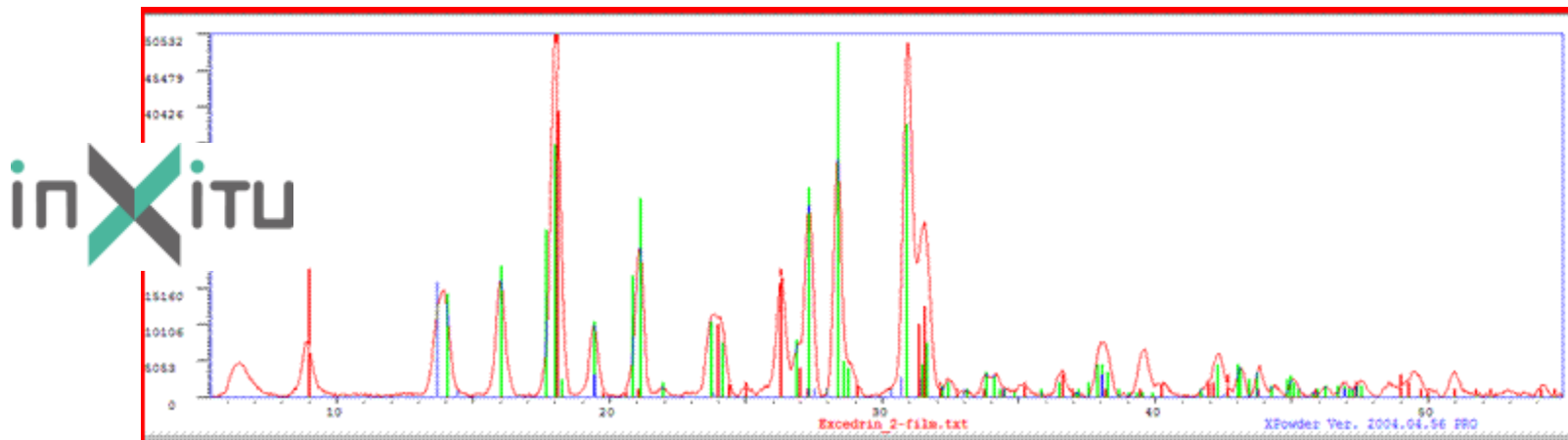




KIT-ANKA PDIFF



ΦΟΡΗΤΟ ΠΕΡΙΘΛΑΣΙΜΕΤΡΟ ΑΚΤΙΝΩΝ-X (XRD)



INSTRUMENTATION FOR GEOLOGICAL FIELD WORK ON THE MOON

D. L. TALBOYS, G. W. FRASER, R. M. AMBROSI, N. NELMS
N. P. BANNISTER, M. R. SIMS, D. PULLAN and J. HOLT
Space Research Centre, University of Leicester, University Road, Leicester, LE2 7RHUK
(*E-mail: dlt3@star.le.ac.uk*)

(Accepted 26 May 2005)

Abstract. A human return to the Moon will require that astronauts are well equipped with instrumentation to aid their investigations during geological field work. Two instruments are described in detail. The first is a portable X-ray Spectrometer, which can provide rapid geochemical analyses of rocks and soils, identify lunar resources and aid selection of samples for return to Earth. The second instrument is the Geological and Radiation environment package (GEORAD). This is an instrument package, mounted on a rover, to perform *in-situ* measurements on the lunar surface. It can be used for bulk geochemical measurements of rocks and soils (particularly identifying KREEP-enriched rocks), prospect for ice in shadowed areas of craters at the poles and characterise the lunar radiation environment.

ΡΟΜΠΟΤΙΚΗ ΟΡΥΚΤΟΛΟΓΙΚΗ ΑΝΑΛΥΣΗ

CHEMIN (CHEMistry-MINeralogy):

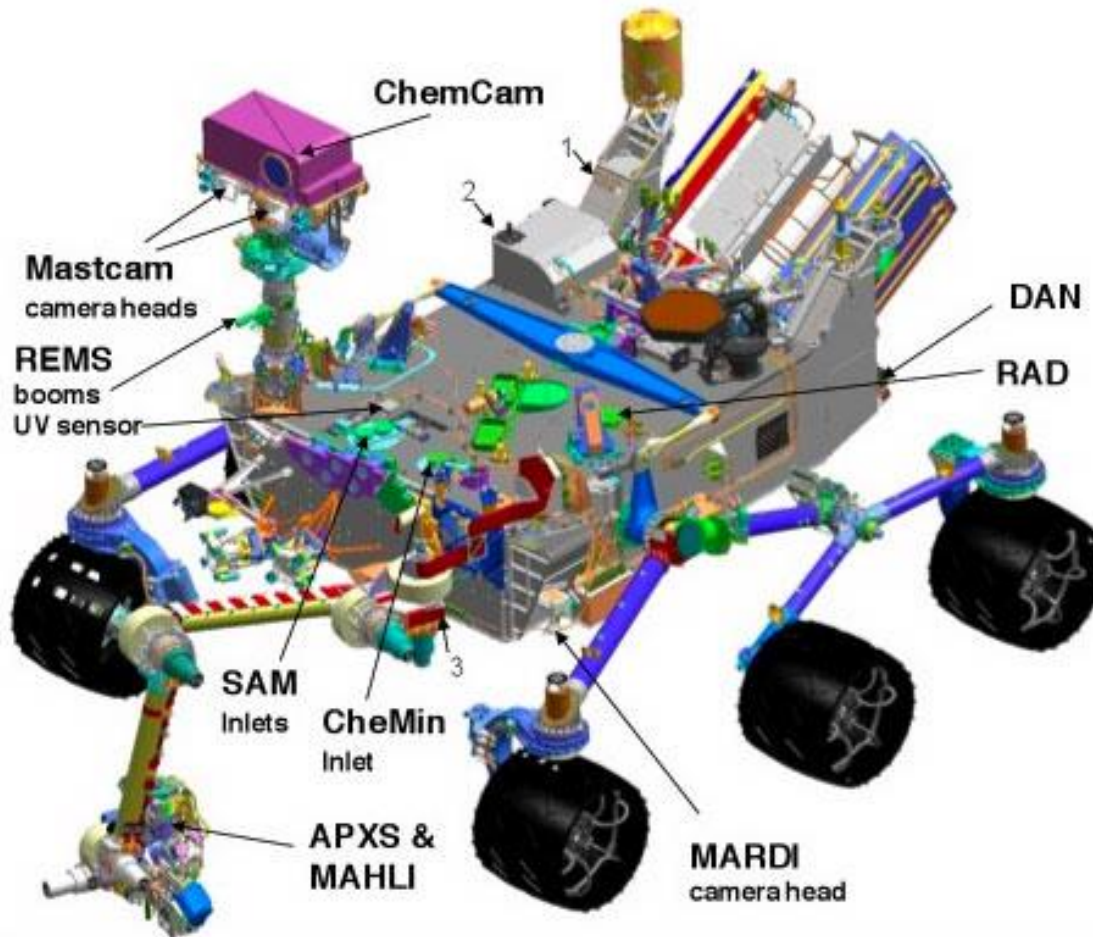
ΦΟΡΗΤΟ ΣΥΣΤΗΜΑ XRD/XRF ΓΙΑ ΤΟ Mars Science Laboratory
(**Curiosity Rover**)



ΡΟΜΠΟΤΙΚΗ ΟΡΥΚΤΟΛΟΓΙΚΗ ΑΝΑΛΥΣΗ

CHEMIN (CHEMistry-MINeralogy):

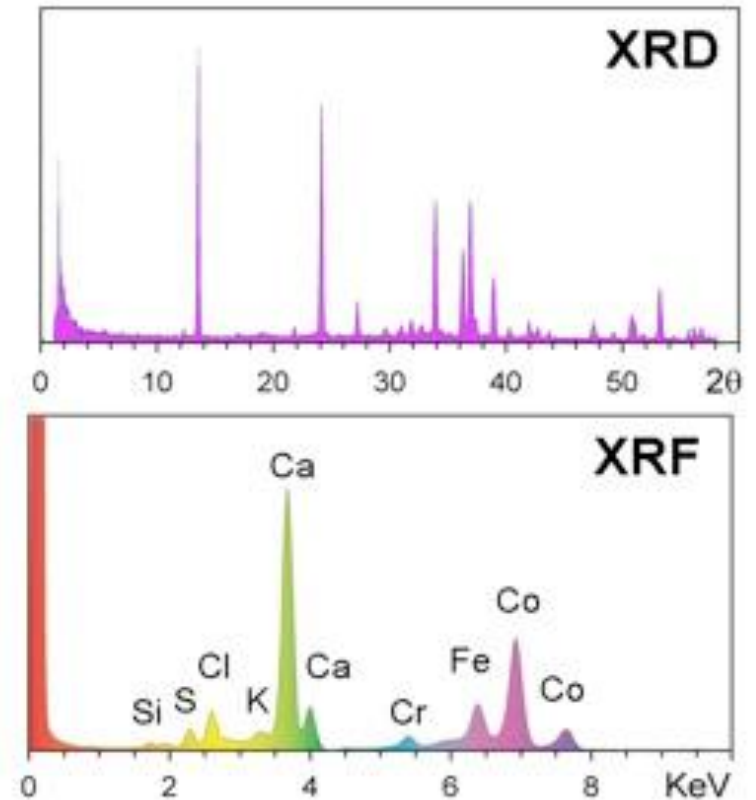
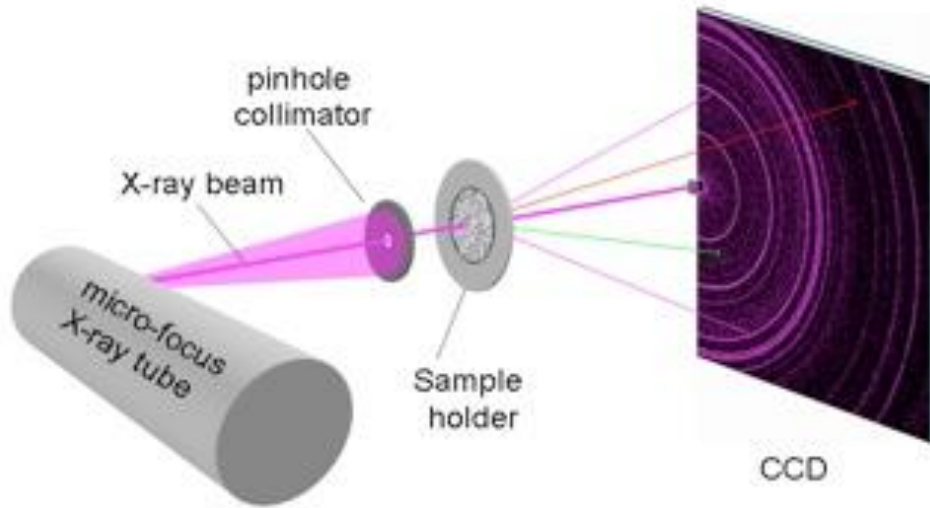
ΦΟΡΗΤΟ ΣΥΣΤΗΜΑ XRD/XRF ΓΙΑ ΤΟ Mars Science Laboratory
(**Curiosity Rover**)



ΡΟΜΠΟΤΙΚΗ ΟΡΥΚΤΟΛΟΓΙΚΗ ΑΝΑΛΥΣΗ

CHEMIN (CHEMistry-MINeralogy):

ΦΟΡΗΤΟ ΣΥΣΤΗΜΑ XRD/XRF ΓΙΑ ΤΟ Mars Science Laboratory (**Curiosity**)



Field deployment of a portable X-ray diffraction/X-ray fluorescence instrument on Mars analog terrain

P. Sarrazin^{a)}

inXitu, P.O. Box 730, Mountain View, California 94042

D. Blake and S. Feldman

NASA Ames Research Center, MS 239-4, Moffett Field, California 94035

S. Chipera and D. Vaniman

Los Alamos National Laboratory, MS D469, Los Alamos, New Mexico 87545

D. Bish

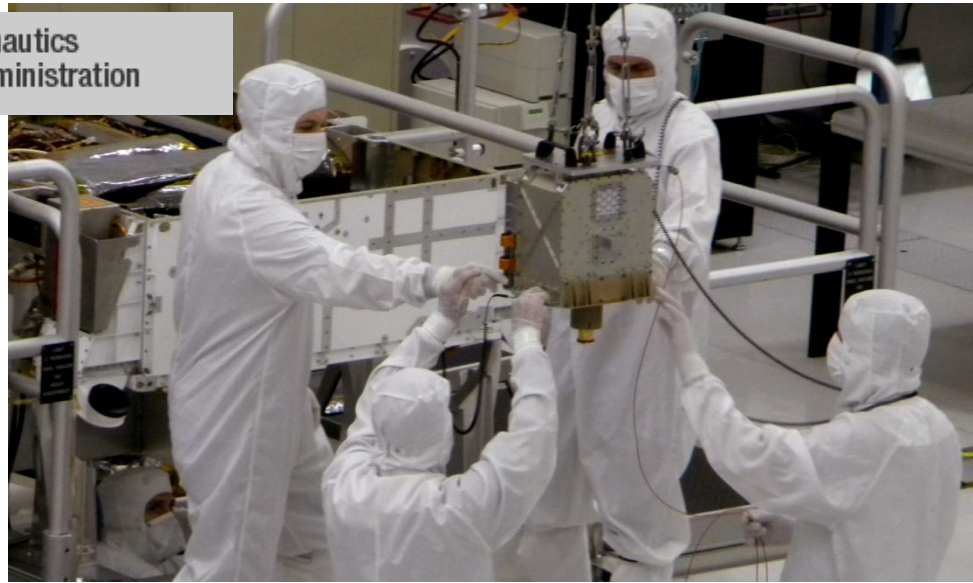
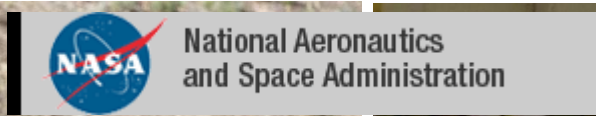
Indiana University, 1001 E 10th Street, Bloomington, Indiana 47405

(Received 31 October 2004; accepted 22 March 2005)

CheMin is a miniature X-ray diffraction/X-ray fluorescence instrument that is included in the payload of the Mars 2009 Mars Science Laboratory mission. A portable CheMin prototype was built to test the capability of the instrument for remote *in situ* mineralogical characterization of geological materials. The instrument was successfully deployed at a variety of Mars analog sites in Death Valley, CA, in May 2004. © 2005 International Centre for Diffraction Data. [DOI: 10.1154/1.1913719]



DAVID BLAKE



PHILIPPE SARRAZIN

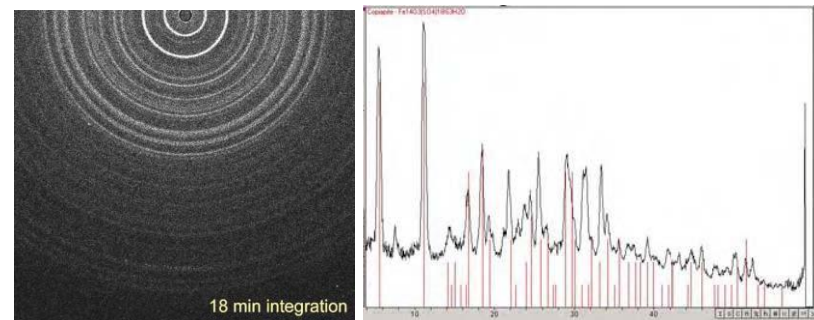


DAVID BISH
Haydn Murray Chair of Applied Clay Mineralogy
 Mineralogy

Office: GY209
 Phone: 812-855-2039
 Email: bish@indiana.edu

Lunar and Planetary Science XXXVIII (2007)

MINERALOGICAL IN-SITU INVESTIGATION OF ACID-SULFATE SAMPLES FROM THE RIO TINTO RIVER, SPAIN, WITH A PORTABLE XRD/XRF INSTRUMENT. P. Sarrazin¹, D.W. Ming², R.V. Morris², D. Fernández-Remolar³, R. Amils³, R.E. Arvidson⁴, D. Blake⁵, D. L. Bish⁶, ¹inXitu Inc., 2551 Casey Ave Ste A, Mountain View, CA 94043 psarrazin@inxitu.com; ²NASA Johnson Space Center, Mail Code KX3, Houston, TX 77058; ³Centro de Astrobiología (CSIC/INTA) 28850 Torrejón de Ardoz, Madrid, Spain; ⁴Earth and Planetary Sciences, Washington University, St. Louis, MO 63130; ⁵NASA Ames Research Center, Moffett Field, CA 94035; ⁶Dept. of Geological Sciences, Indiana Univ., Bloomington, IN 47405-1405.



www.nasa.gov/centers/ames/research/msl_chemin.html

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The Chemistry and Mineralogy Instrument

David Blake, geologist at Ames, is the principal investigator for the The Chemistry and Mineralogy (CheMin) instrument that uses X-ray diffraction - a first for a mission to Mars and a more definitive method for identifying minerals than any instrument on previous missions. It supplements the diffraction measurements with X-ray fluorescence capability to garner further details of composition.

Some minerals detectable by CheMin, such as phosphates, carbonates, sulfates and silica, can help preserve biosignatures. Clay minerals trap and preserve organic compounds under some conditions. Some minerals that form when salty water evaporates can encase and protect organics, too. Other minerals that CheMin could detect might also have implications about past conditions favorable to life and to preservation of biosignatures – or evidence of life.



More Resources:
[Ames' contributions page](#)
[Ames MSL press kit](#)

http://www.nasa.gov/centers/ames/research/msl_chemin.html

Mars Science Laboratory Curiosity Rover

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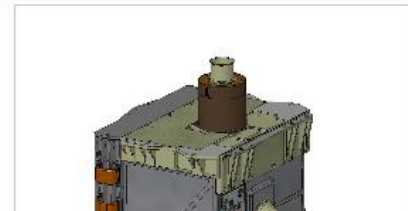
Mission > Instruments > Spectrometers > Chemistry & Mineralogy X-Ray Diffraction (CheMin)

MISSION

ALL INSTRUMENTS	
Cameras:	MastCam MAHLI MARDI
Spectrometers:	APXS ChemCam CheMin SAM
Radiation Detectors:	RAD DAN
Environmental Sensors:	REMS
Atmospheric Sensors:	MEDLI

The Chemistry and Mineralogy instrument, or CheMin for short, will identify and measure the abundances of various minerals on Mars. Examples of minerals found on Mars so far are olivine, pyroxenes, hematite, goethite, and magnetite.

Minerals are indicative of environmental conditions that existed when they formed. For example, olivine and pyroxene, two primary minerals in basalt, form when lava solidifies. Jarosite, found in sedimentary rocks by NASA's Opportunity rover on Mars, precipitates out of water.



<http://mars.jpl.nasa.gov/msl/mission/instruments/spectrometers/chemin/>

ΕΠΙΤΟΠΟΥ (*in-situ*) ΟΡΥΚΤΟΛΟΓΙΚΗ ΑΝΑΛΥΣΗ ΣΤΟΝ ΑΡΗ ΜΕ ΧRD (Οκτώβριος 2012)

www.nasa.gov/mission_pages/msl/index.html

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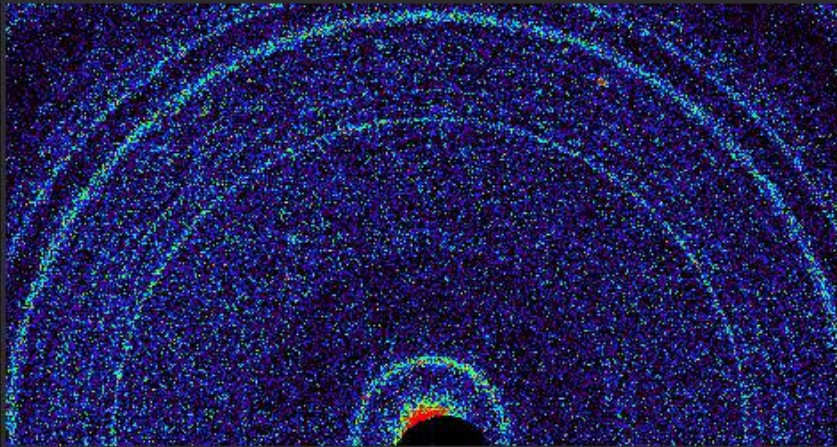


Mars Science Laboratory

Curiosity: Could Mars Have Once Harbored Life?



Latest News



Curiosity Analyzes First Taste of Martian Soil

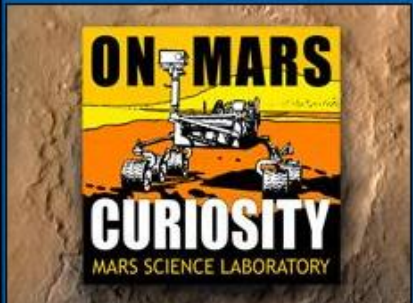
NASA's Mars rover Curiosity has completed initial experiments showing the mineralogy of Martian soil is similar to weathered basaltic soils of ...

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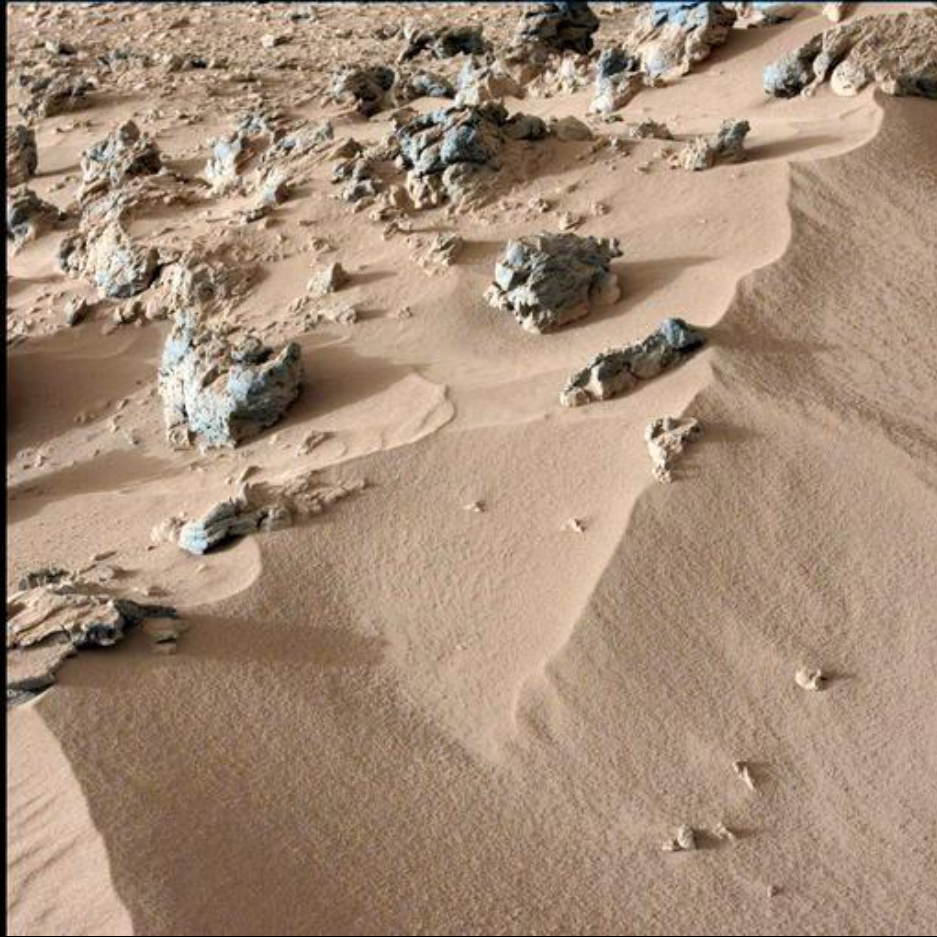
- Working with Curiosity's ChemCam Laser
- Mars Soil Sample Delivered
- Here's the Scoop!



> View This Video

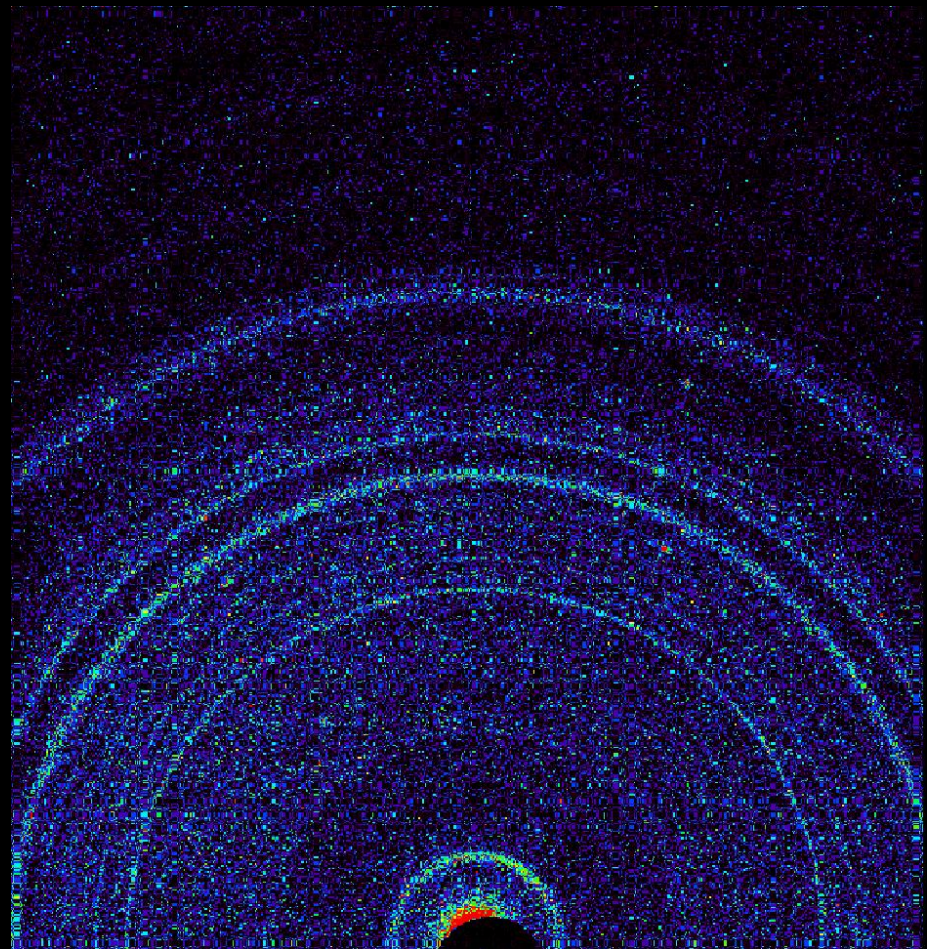
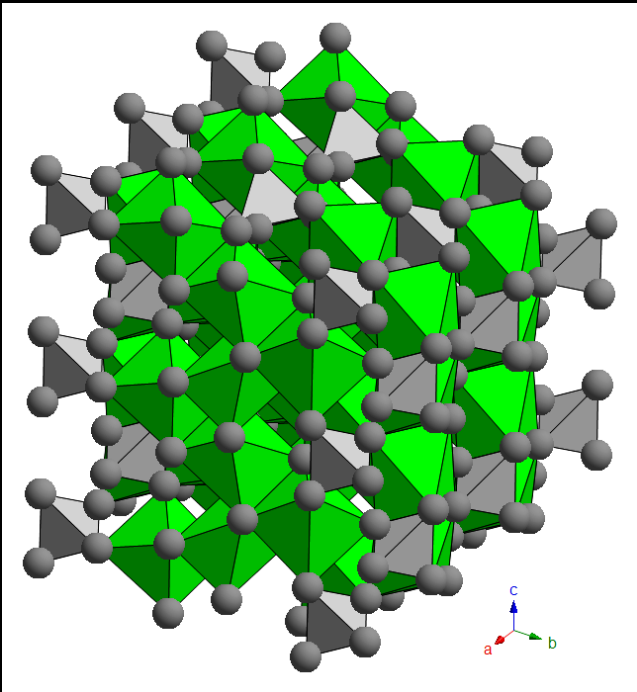
Curiosity's First Scoopful of

ΟΡΥΚΤΑ ΣΤΟ ΕΔΑΦΟΣ ΤΟΥ ΑΡΗ (Οκτώβριος 2012)



ΟΡΥΚΤΑ ΣΤΟ ΕΔΑΦΟΣ ΤΟΥ ΑΡΗ:

- ΟΛΙΒΙΝΗΣ • ΑΣΤΡΙΟΙ • ΠΥΡΟΞΕΝΟΙ



ΦΟΡΗΤΟ ΣΥΣΤΗΜΑ XRD/XRF

in situ XRD/XRF

Specifications

XRD resolution: 0.25° 2θ FWHM

XRD range: 20-55° 2θ

Detector type: 1024 x 256 pixels - 2D Peltier-cooled CCD

XRD geometry: Reflection based

Sample orientation: Flat or convex of infinite size

XRF energy resolution: 200 eV at 5.9 keV

XRF energy range: 3 to 25 keV

X-ray target material: Co or Cu (Cu standard)

X-ray tube voltage: 30kV

X-ray tube power: 10W

Data Storage: 40 Gb - Ruggedized internal hard drive

Wireless Connectivity: 802.11 b/g for remote control from web browser

Operating Temperature: -10°C to 35°C

Dimensions: 48.5 x 39.2 x 19.2cm base; 8.5 X 16 X 4 cm Head

Weight: 12 kg base; 7kg head

About inXitu inc.

inXitu Inc. is a leader in portable XRD/XRF based instrumentation. We specialize in developing the technologies required to enable the next generation of scientific instruments used for materials analysis. The technology used in our portable rock and mineral analyzer received a prestigious R&D 100 award and has been chosen to fly on the Mars Science Laboratory rover scheduled for launch in 2011.



Duetto is featured in the May 2008 issue of Nature magazine.



Duetto

X-ray Diffraction / X-ray Fluorescence instrument for art conservation